

- (b) Why the rate of above reaction decreases on addition of H_2O_2 [3]

UNIT-III

6. (a) Discuss nature of bonding in metal carbonyls. [5]
(b) Giving reason arrange following in order of increasing C-O bond order : [3]
 $[Ni(CO)_4]_d$, $[Co(CO)_4]_d$, $[Fe(CO)_5]_d$
(c) Give reason for the stability of square planar organometallic complexes of Ni (II), although these complexes do not obey $18e^-$ rule. [2]
7. (a) Discuss the rate of reactivity of organoberyllium compounds with N, O, S and P donors. [4]
(b) M-C bond cleavage reaction in Transition metal alkyls. [3]
(c) Why σ -bonded organometallic compounds of transition metals are less stable compared to their main group analogue. [3]

UNIT-IV

8. (a) Discuss microwave spectroscopy and give selection rule for rotational spectra of a molecule [5]

- (b) First line corresponding to $J=0$ appears at 3.84235 cm^{-1} in the rotational spectrum of $^{12}C^{16}O$ molecule (C-12 O-16). Calculate its moment of inertia and bond length. [5]

$$(h = 6.626 \times 10^{-27} \text{ erg/sec}, N = 6.022 \times 10^{23})$$

9. (a) Discuss application of IR spectra in elucidating the structure and properties of polyatomic molecules. [5]
(b) Explain why in IR spectrum of CO_2 molecule, only two absorption bands at 2349 cm^{-1} and 667 cm^{-1} respectively are observed. [5]

OR

Discuss the factors which affect the band position and intensities in I.R. region.

--- x ---

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