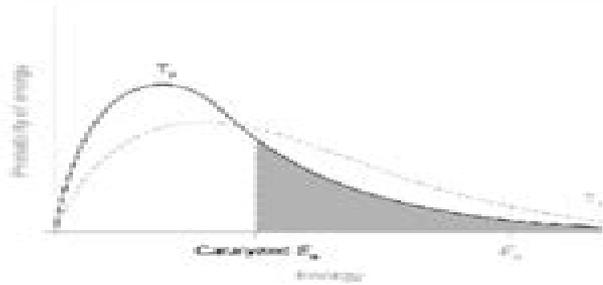


MS Chemistry HL Paper 2 RCA 2023

2.	f	<p>a. <math>Zn^{2+}</math> ions in full shell  <b>OR</b>                      dentic number is zero since no unpaired d orbitals</p>	<p>Do not accept <math>Zn</math> in any transition metal.                      Do not accept structures for <math>Zn^{2+}</math> ions.</p>	1
3.	b	<p>ligands donate pairs of electrons to metal ions  <b>OR</b>                      form coordinate covalent bonds</p> <p>ligands use lone pairs  <b>AND</b>                      metal uses empty d orbitals</p>		1
1.	d	<p>Electrostatic attraction ✓                      Between oppositely charged ions/between <math>Fe^{2+}</math> and <math>SO_4^{2-}</math> ✓</p>		2
4.	a	 <p>same higher <b>AND</b> to left of <math>T_1</math> ✓                      rearranged <math>E_a</math> marked <b>AND</b> to the left of <math>E_a</math> of curve <math>T_1</math> ✓</p>	<p>Do not promote curve into a table, and justify clearly: height and width, or crossing it over after it is.</p> <p>Do not award AP if curve shown shows significantly more than 10% of molecules with insufficient energy since shaded area under curve 1.</p> <p>Accept <math>E_a</math> drawn to <math>T_2</math> instead of curve shown or drawn to left of modified <math>E_a</math>.</p>	2
1.	b	<p>no same <b>AND</b> same <math>T_2</math> ✓</p>		1

3. Equilibrium favours backward direction/shifts to LHS **AND**

As increase in pressure favours side with less moles of gas.

[1]

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