

CHEMISTRY MULTIPLE CHOICE QUESTIONS

A. Atoms, molecules and stoichiometry

2002 -2019



C

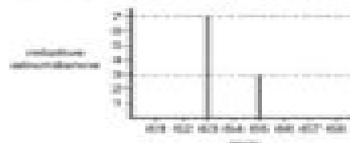
As a combustion, an alkane burns can be considered to have a fully set of 1.00 kg of methane (empirical formula CH_4).

Which mass of carbon dioxide does a person produce each day if all the carbon they take in is digested and oxidised?

- A 0.007 kg B 0.006 kg C 1.30 kg D 0.04 kg

(2002 NA/12)

The diagram shows the mass spectrum of a sample of naturally-occurring copper.



What is the relative atomic mass of this copper?

- A 63.2 B 63.5 C 63.8 D 64.0

(2002 NA/12)

Which pairs of compounds have the same empirical formula?

- ethane and ethene
- ethane and cyclohexane
- cyclohexane and oct-1-ene

(2002 NA/12)

A stream of 10cm^3 of methane and 10cm^3 of ethane was sparked with an excess of oxygen. After cooling to room temperature, the residual gas was passed through aqueous potassium hydroxide.

Which compounds have the empirical formula CH_2O ?

- methanol
- ethanoic acid
- methyl methanoate

(2002 NA/12)

What is the number of molecules in 500 cm^3 of oxygen under room conditions?

- A 1.25×10^{23}
 B 1.25×10^{24}
 C 3.0×10^{23}
 D 3.0×10^{24}

(2004 M/12)

In the preparation of soft margarine, glycerol trioleate



is fully hydrogenated so that, on average, one of its side-chains is converted into the $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ moiety and two side-chains are converted into the $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ moiety.

How many moles of hydrogen are required to convert one mole of glycerol trioleate into the soft margarine?

- A 4 B 5 C 6 D 8

(2008 M/12)

Use of the Data Booklet is relevant to this question.

Chemistry Multiple Choice Questions And Answers Doc

Katrin Zwirgmaier



Chemistry Multiple Choice Questions And Answers Doc:

This book delves into Chemistry Multiple Choice Questions And Answers Doc. Chemistry Multiple Choice Questions And Answers Doc is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemistry Multiple Choice Questions And Answers Doc, encompassing both the fundamentals and more intricate discussions.

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