

## Assessment

**Chapter Test B****Teacher Notes and Answers****8** *Chemical Equations and Reactions***TEST B**

1. b
2. a
3. b
4. d
5. a
6. b
7. d
8. a
9. coefficient
10. liquid
11. reversible reaction
12.  $A + X \rightarrow AX$
13.  $AX \rightarrow A + X$
14.  $A + BX \rightarrow AX + B$
15.  $AX + BY \rightarrow AY + BX$
16. metal hydroxides
17. combustion reaction
18. electrolysis
19. activity series
20. chemical equation
21. conservation of mass
22. gaseous
23. heated
24. A word equation and a formula equation are both qualitative. Although they both show the reactants and products of a chemical reaction, they give no information about amounts of reactants or products. A chemical equation, however, does show the relative amounts of the reactants and products in a chemical reaction.
25. Three indications of a chemical reaction are the radiation of heat and light, the production of a gas, and the formation of a precipitate.
26. zinc sulfide + oxygen  $\rightarrow$   
zinc oxide + sulfur dioxide
27. sodium iodide + chlorine  $\rightarrow$   
sodium chloride + iodine
28.  $Mg + O_2 \rightarrow MgO$
29.  $CaCO_3 \rightarrow CaO + CO_2$
30. decomposition reaction
31. single-displacement reaction
32. synthesis reaction
33. double-displacement reaction
34.  $2AgNO_3 + CuCl_2 \rightarrow 2AgCl + Cu(NO_3)_2$
35.  $2PbO_2 \rightarrow 2PbO + O_2$
36.  $Zn(OH)_2 + 2CH_3COOH \rightarrow$   
 $Zn(CH_3COO)_2 + 2H_2O$
37.  $Mg(s) + 2H_2O(g) \rightarrow Mg(OH)_2(aq) + H_2(g)$
38. no reaction
39.  $Cl_2(g) + MgBr_2(aq) \rightarrow MgCl_2(aq) + Br_2(l)$
40.  $Zn(s) + 2HCl(aq) \rightarrow ZnCl_2(aq) + H_2(g)$
41. no reaction
42.  $H_2(s) + CuCl_2(aq) \rightarrow H_2Cl_2(aq) + Cu(s)$
43. no reaction
44.  $Mg(s) + Co(NO_3)_2(aq) \rightarrow$   
 $Mg(NO_3)_2(aq) + Co(s)$

# Holt Chemistry Chapter 8 Test

**K Morrison**



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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Holt Chemistry Chapter 8 Test**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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