




Introduction to Solutions

Instructions: Circle the substances below that would be classified as **solutions**.

<p>Apple Juice</p> 	<p>Salad</p> 	<p>Dirt</p> 	<p>Lemonade</p> 
<p>Coffee with Sugar</p> 	<p>Pure Water (H₂O)</p> 	<p>Cookie</p> 	<p>Soda Pop</p> 

Instructions: Use the choices in the word bank below to answer the following questions:

Aqueous	Electrolyte	Solution	Water	Solute	Solvent	Solvation
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- 1) Solute The substance that is dissolved in a solution (i.e. sugar in water)
- 2) Solution A homogenous mixture of two or more substances.
- 3) Electrolyte Solution that can conduct an electric current due to dissolved ions.
- 4) Solvation The process of dissolving. Solute particles are surrounded by solvent.
- 5) Aqueous A solution in which the solvent is water (H₂O)
- 6) Solvent The substance that is doing the dissolving in a solution (i.e. water)
- 7) Water This substance is often referred to as the "universal solvent."

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