

## Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. The charge on an ion is  
a. always positive.  
b. always negative.  
c. either positive or negative.  
d. zero.
- \_\_\_\_\_ 2. According to the octet rule, a calcium atom has a tendency to  
a. lose one electron.  
b. lose two electrons.  
c. gain one electron.  
d. gain two electrons.
- \_\_\_\_\_ 3. If a compound forms by ionic bonding, which is *not* true?  
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.  
b. The net charge of the compound is zero.  
c. The compound contains just two atoms, each of opposite charge.  
d. Several ions group together in a tightly packed structure.
- \_\_\_\_\_ 4. The only property listed that is *not* characteristic of ionic compounds is  
a. high melting point.  
b. hardness.  
c. lack of crystal structure.  
d. good conductor of electricity.
- \_\_\_\_\_ 5. Which formula listed below represents a polyatomic ion?  
a.  $\text{NH}_4^+$   
b.  $\text{H}_2\text{SO}_4$   
c.  $\text{NaCl}$   
d.  $\text{H}_2\text{O}$
- \_\_\_\_\_ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because  
a. ionic substances tend to vaporize at room temperature.  
b. ionic substances are brittle.  
c. attractive forces between ions are greater than the attractive forces between molecules.  
d. the numbers of positive and negative charges are equal in an ionic compound.

# Chapter 6 Chemical Bonding Test

**LP Steffe**



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, Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **Materials Laboratory Procedures Manual** United States. Bureau of Reclamation, 1957 **McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition** Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill s SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need perform at your best This book includes 4 full length sample tests updated for the latest test format 40 top tips to remember on test day Glossary of tested chemistry terms and formulas Tips and strategies from one of the most popular teachers at the renowned Brooklyn Technical High School Diagnostic test to pinpoint strengths and weaknesses Step by step review of all topics covered on the exam In depth coverage of the lab experiment questions that are a major test feature Charts tables and illustrations to simplify and reinforce learning Practice tests just like the real SAT Subject Test in Chemistry Test taking tips and strategies **McGraw-Hill's SAT Subject Test: Chemistry, 2ed** Thomas Evangelist, 2009-02-01 We want to help you score high on the SAT Chemistry test We ve put all of our proven expertise into McGraw Hill s SAT Subject Test Chemistry to make sure you re fully prepared for this difficult exam With this book you ll get essential skill building techniques and strategies created by a leading high school chemistry teacher You ll also get 5 full length practice tests hundreds of sample questions and all the facts about the current exam With McGraw Hill s SAT Subject Test Chemistry we ll guide you step by step through your preparation program and give you the tools you need to succeed 5 full length sample tests with complete explanations for every question 40 top test items to remember on exam day A step by step review of all topics covered on the exam Teacher recommended tips and strategies to help you raise your score **Instructor's Manual and Test Bank to Accompany Basic Concepts of Chemistry** Malone, 2000-04-18 **Chemistry 2007** David Wilson, 2006-12 Kaplan s guide includes 2 full length practice tests Diagnostic test to target areas for score improvement Detailed answer explanations Hundreds of practice questions from calculations of chemical equations to organic chemistry Explanations of important terms formulas and concepts Powerful strategies to help you score higher Radiochemical Analysis of Agrochemicals Mahbubul A. F. Jalal, Svetlana Bondarenko, 2025-03-06 This book provides an in depth overview of agrochemicals their development regulation and environmental impact It also offers detailed insights into the latest research methodologies and regulatory standards Through this comprehensive guide readers will gain a thorough understanding of the complexities surrounding agrochemical

use and its implications for modern agriculture The chapters cover a wide range of topics including the history and evolution of crop protection methods the regulatory framework governing agrochemicals and the intricate processes of degradation and metabolism in various environments Expert contributors discuss the nuances of radioactivity in agrochemical studies quantitative and spectroscopic analysis techniques and the behaviour of these chemicals in soil water plants and livestock Given its breadth this book appeals to academics practitioners policymakers and students interested in understanding the scientific principles and practical applications of agrochemicals Researchers in the fields of agricultural science

environmental chemistry and regulatory affairs will find this book invaluable **Chemogenomics and Chemical Genetics**

ERIC MARECHAL,Sylvaine Roy,Laurence Lafanechère,2011-06-21 Biological and chemical sciences have undergone an unprecedented transformation reflected by the huge use of parallel and automated technologies in key fields such as genome sequencing DNA chips nanoscale functional biology or combinatorial chemistry It is now possible to generate and store from tens of thousands to millions of new small molecules based on enhanced chemical synthesis strategies Automated screening of small molecules is one of the technologies that has revolutionized biology first developed for the pharmaceutical industry and recently introduced in academic laboratories High throughput and high content screening allow the identification of bioactive compounds in collections of molecules chemical libraries being effective on biological targets defined at various organisational scales from proteins to cells to complete organisms These bioactive molecules can be therapeutic drug candidates molecules for biotech diagnostic or agronomic applications or tools for basic research Handling a large number of biological genomic and post genomic chemical and experimental information screening approaches cannot be envisaged without any electronic storage and mathematical treatment of the data Chemogenomics and Chemical Genetics is an introductory manual presenting methods and concepts making up the basis for this recent discipline This book is dedicated to biologists chemists and computer scientist beginners It is organized in brief illustrated chapters with practical examples Clear definitions of biological chemical and IT concepts are given in a glossary section to help readers who are not familiar with one of these disciplines Chemogenomics and Chemical Genetics should therefore be helpful for students from Bachelor s degree level technological platform engineers and researchers in biology chemistry bioinformatics cheminformatics both in biotech and academic laboratories

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