

Assessment 3

Chapter Test B**Chapter: Chemical Equations and Reactions**

Part 1 In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The production of a slightly soluble solid compound by a double displacement reaction is called the formation of a
- gas.
 - precipitate.
 - redox reaction.
 - halogen.
- _____ 2. To balance a chemical equation, it is permissible to adjust the
- coefficients.
 - subscripts.
 - formula of the products.
 - number of products.
- _____ 3. In a chemical equation, the symbol (aq) indicates that the substance is
- solid.
 - dissolved in water.
 - an acid.
 - molten.
- _____ 4. The tendency for a replacement reaction to occur depends on the
- activity between any two elements in the activity series themselves.
 - temperature difference.
 - valence electrons and mass of.
 - activity between any two elements in the activity series themselves.
- _____ 5. The coefficients in a chemical equation
- indicate the number of moles of each substance that react.
 - show the number of grams of each substance that react.
 - are the molar masses of the substances.
 - show the valence electrons that each atom.
- _____ 6. If metal X is more reactive than metal Y in the activity series, then
- X will replace ions of Y in solution.
 - Y will replace ions of X in solution.
 - Y will form oxides only indirectly.
 - X will react with cold water.

Chemistry Chapter 8 Study Guide Answers

Michael Seilmaier



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Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 *Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition* Susan McMurry, 2000 Provides answers and explanations to all in text and end of chapter exercises Also includes summaries of name reactions functional group synthesis and reactions lists of reagents and abbreviations and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry This edition now includes all new artwork expanded in text problems summary quizzes approximately every three chapters more detailed explanations in solutions and chapter outlines **Study Guide and Solutions Manual for Organic Chemistry** Susan McMurry, John McMurry, 1988 **Study Guide and Solutions Manual to Accompany The Essence of Organic Chemistry** Jane M. Cram, Donald J. Cram, 1978 Study Guide and Solutions Book Harold Hart, 1983 **Organic Chemistry Study Guide with Solutions Manual** Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer **Study Guide and Solutions Manual for Fundamentals of Organic Chemistry** Susan McMurry, 1994 Following a brief review of structure and bonding organic molecules and functional groups are presented as early as possible The text is organized primarily by functional group beginning with simple alkanes and moving toward more complex compounds Emphasis is placed on the fundamental mechanistic similarities of organic reactions McMurry's thorough revision continues to present the solid content necessary for this course without sacrifice of important subjects and pedagogical tools Text and reaction summaries full problem sets and outstanding artwork are just some of the features in the Third Edition usually found in a full year book McMurry's clear well written explanations remain a highlight of the book *Study guide to accompany chemistry* Karen Timberlakes, 1983 **Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e** Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry *Study Guide with Solutions to Accompany Burgoyne, A Study Course in Organic Chemistry* N. Sue Barnes, 1979 **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** John McMurry, Susan McMurry, 1986 *Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry* McMurry, Susan McMurry, 1990 **Student Study Guide Fundamentals of Chemistry by David E. Goldberg** Margaret G. Kimble, 1994 Study Guide, Answer Book to Accompany Organic Chemistry Alan S. Wingrove, Robert L. Caret, 1981 *Instructor's Manual and Study Guide Answers for the Human Body in Health and Disease* Barbara Janson Cohen, Memmler, 1996

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