

CHAPTER # 1

STOICHIOMETRY

Q1. Define stoichiometry.

Ans: Stoichiometry:

The study of relative amounts of substances involved in a chemical reaction is called Stoichiometry. Such phenomenon is studied through the knowledge of Stoichiometry (Greek word *Stolcheion* means element and *metry* means measurement).

Importance of Stoichiometry:

Stoichiometry is essential when quantitative information about a chemical reaction is required. Moreover it is important to predict yields of chemical products.

Q2. Explain the significance of stoichiometry with the help of example.

OR

How will you explain law of conservation of mass in the case of combustion of hydrogen fuel in rockets?

Ans: Combustion of hydrogen fuel in rockets:

1. Consider a rocket manufacturer uses liquid hydrogen as a fuel. He may have to determine how much fuel is necessary for a particular flight. Hydrogen burns in oxygen (of air) to produce water.



It states that

- (i) Two moles of Hydrogen react with one mole of oxygen to form two moles of steam.
- (ii) Two molecules of Hydrogen react with one molecule of oxygen to produce two molecules of steam.
- (iii) Four grams of hydrogen react with thirty-two grams of oxygen to produce thirty-six grams of water. Here the total mass of reactants is equal to the total mass of products. Thus it confirms the Law of conservation of mass.
2. Another example is the reaction taking place in a gas barbecue. This is the example of combustion to form carbon dioxide and water. The balanced chemical reaction is



Q3. What do you understand by the term Mole?

Ans: Mole:

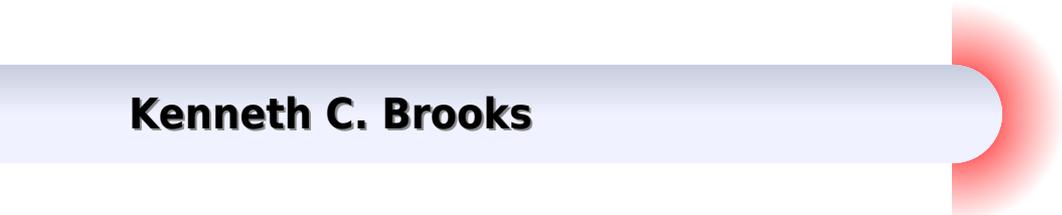
The atomic mass, formula mass and molecular mass of a substance expressed in grams is called Mole.

$$\text{Number of moles} = \frac{\text{mass in grams}}{\text{molecular mass}}$$

Example: One mole of O = 16 g

Chemistry Chapter 9 Stoichiometry

Kenneth C. Brooks



Chemistry Chapter 9 Stoichiometry:

The IIT Foundation Series - Chemistry Class 8, 2/e, Holt Chemistry Ralph Thomas Myers, 2004 Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2000-12 *Fundamentals of Chemistry* Fred H. Redmore, 1979

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web use the dichotomous key provided to identify the salamanders in figure 1 write the pathway you took to get to the name of the salamander next to the drawing write the correct name for the salamander on the line below each picture figure 1 types of salamanders key to the salamanders a hind limbs absent hind limbs present

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web procedure use the dichotomous key provided on the back of this sheet to identify at least 3 species of salamanders in addition to the two we will work through as a class begin by reading statements 1a and 1b one of the statements describes the salamander the other statement does not

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web use the following dichotomous key to correctly identify the species of salamanders designated in the pictures place the name of the salamander beside the number on the answer sheet classification key for certain salamanders 1 a hind limbs absent siren intermedia siren b hind limbs present go to 2 2 a

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web 1 examine the drawings of the salamanders in figure 1 below you and your group must work together to identify each salamander using the key on pg 3 figure 2 2 use the dichotomous key below figure 2 to determine the genus and species of each salamander step 1 read statements 1a and 1b

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web the reader studies the specimen and selects the descriptions that apply to it until reaching a statement that characterizes only one species and names it in this investigation you will use a typical dichotomous key to identify the genus and

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web 10 q a 3 more from salamander science description i use this activity or short lab to allow students to practice using dichotomous keys it also gives a great opportunity to review or introduce taxonomy and classification particularly why scientific names are important

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web dichotomous key to the salamanders 1 a hind limbs absent siren b hind limbs present go to 2 2 a external gills present in adults mud puppy b external gills absent in adults go to 3 3 a large size over 7 cm long go to 4 small size under 7 cm long go to 5 body background black large white spots irregular in shape

