

STUDY GUIDE

SECTION 1.1 - 1.4

- **matter**: everything that takes up space + has a mass

Properties of matter



① Physical changes:

- Phase
- Size
- Texture

→ Reversible

② Chemical changes:

- Color change
- New substance is formed

→ Irreversible

Pure substance vs. Mixture

MOLECULE (pure substances)

Compounds

2 or more ≠ atoms
(eg: H_2O ; Na_2O)

Elements

1 atom
(eg: H_2 , O_2)

- **Pure substance**: constant composition + definite properties

Chemistry Midterm Study Guide

RS Peters



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