

1. Which condition makes for a better solution to this unconstrained problem than (a)?
  - (a)  $f$  is a symmetric function that is concave down, the average starting energy per molecule of the hydrogen molecule component decreases steadily as the reaction proceeds, and the reaction is reversible.
  - (b)  $f$  is a concave down function of energy,  $f(0) = 0$ ,  $f(1) = 1$ ,  $f(2) = 2$ ,  $f(3) = 3$ ,  $f(4) = 4$ ,  $f(5) = 5$ .
  - (c)  $f$  is a concave down function of energy,  $f(0) = 0$ ,  $f(1) = 1$ ,  $f(2) = 2$ ,  $f(3) = 3$ ,  $f(4) = 4$ ,  $f(5) = 5$ .
  - (d)  $f$  is a concave down function of energy,  $f(0) = 0$ ,  $f(1) = 1$ ,  $f(2) = 2$ ,  $f(3) = 3$ ,  $f(4) = 4$ ,  $f(5) = 5$ .
2. (a) How many units of hydrogen molecule are there? (b) How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000
  - (c) 1000 (d) 1000
3. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
4. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
5. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
6. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
7. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
8. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
9. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
10. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
11. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
12. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
13. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
14. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000
15. The total energy of the system is 1000 units. How many units of hydrogen molecule are there?
  - (a) 1000 (b) 1000 (c) 1000 (d) 1000

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**Christopher Bruhn**



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