



COVALENT BONDING

Practice Problems

Do your answers to the following problems.

SECTION 8.1 MOLECULAR COMPOUNDS

- Classify each of the following as either ionic or covalent.
 - NaCl
 - H_2O
 - HCl
 - KNO_3
 - H_2SO_4
 - H_2O_2
 - H_2
 - H_2S
 - H_2O

- Classify each of the following as either ionic or covalent compound. Name it.
 - K_2SO_4
 - H_2O
 - H_2SO_4
 - H_2O_2
 - H_2S
 - H_2O
 - H_2S
 - H_2O
 - H_2S

SECTION 8.2 THE NATURE OF COVALENT BONDING

- Show covalent bonding in hydrogen fluoride. $\text{H}-\text{F}$
- Show covalent bonding in methane. C with four H atoms around it.
- Show covalent bonding in carbon dioxide. C with two O atoms around it.
- Show covalent bonding in ethane. $\text{H}-\text{C}-\text{C}-\text{H}$ with appropriate H atoms.
- Show covalent bonding in acetylene. $\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw Lewis structures for the following molecules.
 - H_2O
 - H_2S
 - H_2O_2
 - H_2SO_4
 - H_2SO_3
 - H_2SO_2
 - H_2SO
 - $\text{H}_2\text{S}_2\text{O}_7$
 - $\text{H}_2\text{S}_2\text{O}_8$
 - $\text{H}_2\text{S}_2\text{O}_9$
 - $\text{H}_2\text{S}_2\text{O}_{10}$
 - $\text{H}_2\text{S}_2\text{O}_{11}$
 - $\text{H}_2\text{S}_2\text{O}_{12}$
 - $\text{H}_2\text{S}_2\text{O}_{13}$
 - $\text{H}_2\text{S}_2\text{O}_{14}$
 - $\text{H}_2\text{S}_2\text{O}_{15}$
 - $\text{H}_2\text{S}_2\text{O}_{16}$
 - $\text{H}_2\text{S}_2\text{O}_{17}$
 - $\text{H}_2\text{S}_2\text{O}_{18}$
 - $\text{H}_2\text{S}_2\text{O}_{19}$
 - $\text{H}_2\text{S}_2\text{O}_{20}$
 - $\text{H}_2\text{S}_2\text{O}_{21}$
 - $\text{H}_2\text{S}_2\text{O}_{22}$
 - $\text{H}_2\text{S}_2\text{O}_{23}$
 - $\text{H}_2\text{S}_2\text{O}_{24}$
 - $\text{H}_2\text{S}_2\text{O}_{25}$
 - $\text{H}_2\text{S}_2\text{O}_{26}$
 - $\text{H}_2\text{S}_2\text{O}_{27}$
 - $\text{H}_2\text{S}_2\text{O}_{28}$
 - $\text{H}_2\text{S}_2\text{O}_{29}$
 - $\text{H}_2\text{S}_2\text{O}_{30}$
 - $\text{H}_2\text{S}_2\text{O}_{31}$
 - $\text{H}_2\text{S}_2\text{O}_{32}$
 - $\text{H}_2\text{S}_2\text{O}_{33}$
 - $\text{H}_2\text{S}_2\text{O}_{34}$
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 - $\text{H}_2\text{S}_2\text{O}_{95}$
 - $\text{H}_2\text{S}_2\text{O}_{96}$
 - $\text{H}_2\text{S}_2\text{O}_{97}$
 - $\text{H}_2\text{S}_2\text{O}_{98}$
 - $\text{H}_2\text{S}_2\text{O}_{99}$
 - $\text{H}_2\text{S}_2\text{O}_{100}$

SECTION 8.3 BONDING: THE ORBITALS

- Which orbital and sub-orbital are represented by p_x , p_y , p_z , d_{xy} , d_{yz} , d_{zx} , $\text{d}_{x^2-y^2}$, and d_{z^2} ?
- Which orbital and sub-orbital are represented by p_x , p_y , p_z , d_{xy} , d_{yz} , d_{zx} , $\text{d}_{x^2-y^2}$, and d_{z^2} ?

Chapter 8 Covalent Bonding Practice Problems Answers

Julie R. Frentrup



Chapter 8 Covalent Bonding Practice Problems Answers:

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Understanding Chemistry Fred M. Dewey,1994

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Organic Chemistry T. W. Graham Solomons,Craig B. Fryhle,Scott A. Snyder,2022-05-26 Organic Chemistry 13th Edition continues Solomons Fryle and Snyder s tradition of excellence in teaching and preparing students for success in both the classroom and beyond Central to the authors is their approach in emphasizing organic chemistry s relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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Chapter 8 Covalent Bonding Practice Problems Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Chapter 8 Covalent Bonding Practice Problems Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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