

2019

Name: Phyllis Ann Smith LAB NUMBER: 200

Suppose that you had the correct apparatus for the following experiment. In what state would the following gases be present?

CO ₂ (g)	H ₂ O(g)	H ₂ (g)	H ₂ (l)
<u>gaseous</u>	<u>condensed</u>	<u>gaseous</u>	<u>liquid</u>

1. If you have the correct apparatus and a 1.00 g sample of H₂ is completely oxidized by combustion of water is produced. How much water is produced?

$$\begin{array}{r}
 1.00 \text{ g H}_2 \times \frac{1 \text{ mole H}_2}{2.016 \text{ g H}_2} \times \frac{2 \text{ moles H}_2\text{O}}{1 \text{ mole H}_2} \times \frac{18.015 \text{ g H}_2\text{O}}{1 \text{ mole H}_2\text{O}} = 9.00 \text{ g H}_2\text{O}
 \end{array}$$

2. If you have the correct apparatus and quantity and a sample of 1.00 g of H₂ is completely oxidized by combustion of water is produced.

$$\begin{array}{r}
 1.00 \text{ g H}_2 \times \frac{1 \text{ mole H}_2}{2.016 \text{ g H}_2} \times \frac{2 \text{ moles H}_2\text{O}}{1 \text{ mole H}_2} \times \frac{18.015 \text{ g H}_2\text{O}}{1 \text{ mole H}_2\text{O}} = 9.00 \text{ g H}_2\text{O}
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 \end{array}$$

9.00 g
9.00 g
9.00 g

Chem 110 Lab Questions And Answers

Henry Walter Young



Chem 110 Lab Questions And Answers:

General Chemistry ... Laboratory Manual and Note Book American School (Lansing, Ill.),1905 **Fundamentals of Chemistry: Laboratory Studies** Frank Brescia,2012-12-02 *Fundamentals of Chemistry Laboratory Studies* Third Edition is a manual that provides instruction on techniques of chemical laboratory operations Each experiment is discussed in terms of the major objective the experimental approach to the objective the measurements or observations to be made and the calculation and interpretation of results Topics covered include manipulation weights and measures molecular weight acids and bases gravimetric and volumetric stoichiometry and thermochemistry This book is comprised of 43 chapters divided into 14 sections and begins by presenting general information on metric and other units common laboratory equipment and chemical laboratory methods The first chapter introduces the reader to the Bunsen burner and the principles of glass working followed by a discussion on mass and volume measurements including the determination of density The following chapters focus on states of matter molecular weight stoichiometry and intermolecular forces Preparations and syntheses are also considered along with chemical equilibrium and electrochemistry The final section is devoted to qualitative analysis particularly of cations and anions This monograph is intended primarily for students of chemistry *Examination Questions in Biology, Botany, Chemistry, Drawing, Geography, Physics, Zoölogy* College Entrance Examination Board,1920

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Popular Electricity and the World's Advocate ,1909 **Popular Electricity and the World's Advance** Henry Walter Young,1909 **Chemical Age** ,1912

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