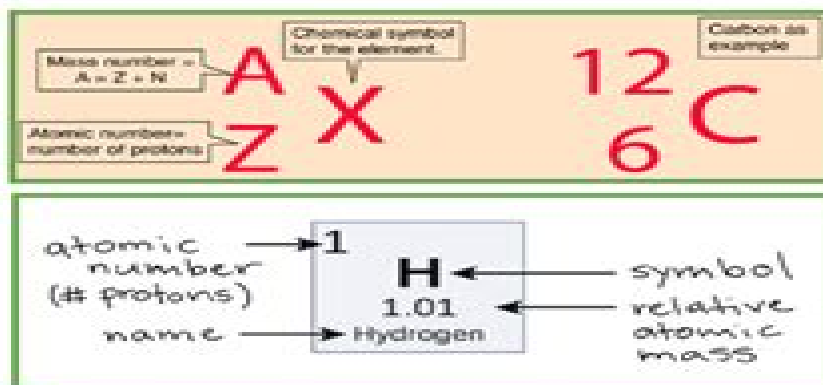


Chapter 1

You should be able to:

- Identify the key steps in the Scientific Method.
 - Step 1: Make observations.
 - Step 2: Formulate a hypothesis.
 - Step 3: Test the hypothesis through experimentation.
 - Step 4: Accept or modify the hypothesis. (develop a model)
 - Step 5: Develop into a law and/or a theory. (conduct further experiments)
- Explain the difference between an atom and an element.
 - **Atom:** is the **smallest component of an element**, containing neutrons, protons and electrons, and **makes up everything around us**.
 - **Element:** is a substance in which all of the atoms have the same atomic (proton) number. It is also defined as a substance which cannot be broken down by chemical means
- Describe the subatomic particles (protons, neutrons and electrons) and their location in the nuclear model of the atom.
 - **Electron:** A **negatively** charged subatomic particle **found outside** the nucleus of all atoms.
 - **Neutrons:** An **electrically neutral** with a mass & diameter slightly greater than that of a proton **located in the nucleus**
 - **Proton:** A **positively** charged particle **in the nucleus** of an atom.
- Identify an element by its atomic number and chemical symbol



- Write the full atomic symbol for any neutral atom or ion.

Chemistry 101 Study Guide

N Noddings



Chemistry 101 Study Guide:

The book delves into Chemistry 101 Study Guide. Chemistry 101 Study Guide is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chemistry 101 Study Guide, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 4: Chemistry 101 Study Guide in Specific Contexts
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