



## REACTING MASS CALCULATIONS - INTRODUCTION

- Step 1** Write ✓ for the substance whose mass is given and ? for the substance whose mass is to be calculated on the balanced equation
- Step 2** Find the moles of the ✓ substance ( using  $\text{moles} = \frac{\text{mass}}{M_r}$  )
- Step 3** Use the balanced equation and your answer from step 2 to find the moles of the ? substance
- Step 4** Find the mass of the ? substance ( using  $\text{mass} = M_r \times \text{moles}$  )

1) What mass of oxygen reacts with 12 g of magnesium?



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2) What mass of calcium hydroxide is made from 14 kg of calcium oxide?



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3) What mass of aluminium is needed to react with 640 g of iron oxide?



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4) What mass of titanium chloride reacts with 460 g of sodium?



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# Calculations In Chemistry An Introduction

**A Gutmann**



## **Calculations In Chemistry An Introduction:**

## Decoding **Calculations In Chemistry An Introduction**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Calculations In Chemistry An Introduction**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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