



7. What is the maximum mass of aluminum hydroxide that can be produced when 25.0 mL of 0.248 M sodium hydroxide solution are mixed with 25.0 mL of 0.189 M aluminum nitrate solution?

Find mol of each reactant

$$0.0250 \text{ L} \times 0.248 \frac{\text{mol}}{\text{L}} = 6.2 \times 10^{-3} \text{ mol NaOH}$$

$$0.0250 \text{ L} \times 0.189 \frac{\text{mol}}{\text{L}} = 4.73 \times 10^{-3} \text{ mol Al}(\text{NO}_3)_3$$

$$6.20 \times 10^{-3} \text{ mol NaOH} \times \frac{1 \text{ mol Al}(\text{OH})_3}{3 \text{ mol NaOH}} \times \frac{78.0 \text{ g Al}(\text{OH})_3}{\text{mol}} =$$

$$\boxed{0.161 \text{ g Al}(\text{OH})_3}$$

need 3 times as much NaOH as $\text{Al}(\text{NO}_3)_3$ but we only have 1.3 x as much NaOH is limiting

8. Indicate which of the following are redox reactions. For those that are redox reactions, label the reactant that is the oxidizing agent and the reactant that is the reducing agent. You do not have to balance these reactions!



Redox agent: C_2H_6 (red.), O_2 (ox.)

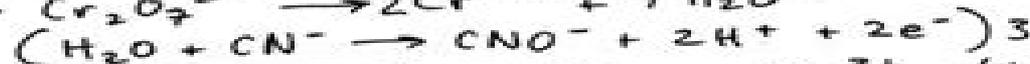


Not redox

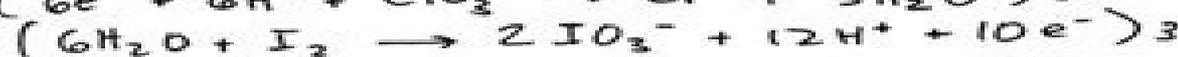


Redox agent: PCl_5 (red.), Cl_2 (ox.)

9. Aqueous dichromate ion reacts with aqueous cyanide ion (CN^-) to produce chromium (III) ion and aqueous cyanate (CNO^-) ion. Balance the equation using the half-reaction method, assuming the reaction occurs in acidic solution.



10. Balance the following redox reaction in basic solution.



Answers To Modern Chemistry Homework Chapter5

M Walker



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Fundamentals of Chemistry: A Modern Introduction (1966) Frank Brescia, 2012-12-02 Fundamentals of Chemistry A Modern Introduction focuses on the formulas processes and methodologies used in the study of chemistry The book first looks at general and historical remarks definitions of chemical terms and the classification of matter and states of aggregation The text then discusses gases Ideal gases pressure of a gas confined by a liquid Avogadro s Law and Graham s Law are described The book also discusses aggregated states of matter atoms and molecules chemical equations and arithmetic thermochemistry and chemical periodicity The text also highlights the electronic structures of atoms Quantization of electricity spectra of elements quantization of the energy of an electron associated with nucleus the Rutherford Bohr nuclear theory hydrogen atom and representation of the shapes of atomic orbitals are explained The text also highlights the types of chemical bonds hydrocarbons and their derivatives intermolecular forces solutions and chemical equilibrium The book focuses as well on ionic solutions galvanic cells and acids and bases It also discusses the structure and basicity of hydrides and oxides The reactivity of hydrides charge of dispersal and basicity effect of anionic charge inductive effect and basicity and preparation of acids are described The book is a good source of information for readers wanting to study chemistry

Children's Books in Print R R Bowker Publishing, Bowker, 1999-12

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Secondary Teaching Methods Kenneth T. Henson, 1981 *Physical Chemistry* Thomas Engel, 2006 *Proceedings of the 2024 9th International Conference on Modern Management, Education and Social Sciences (MMET 2024)* Chunyi Lin, Fong Peng Chew, Intakhab Alam Khan, Catalin Popescu, 2024-12-07 This is an open access book As a leading role in the global megatrend of scientific innovation China has been creating a more and more open environment for scientific innovation increasing the depth and breadth of academic cooperation and building a community of innovation that benefits all Such endeavors are making new contributions to the globalization and creating a community of shared future To adapt to this changing world and China's fast development in the new era 2024 9th International Conference on Modern Management Education and Social Sciences MMET 2024 to be held in September 20 22 2024 This conference takes bringing together global wisdom in scientific innovation to promote high quality development as the theme and focuses on cutting edge research fields including Modern Management and Education Technology MMET 2023 encourages the exchange of information at the forefront of research in different fields connects the most advanced academic resources in China and the world transforms research results into industrial solutions and brings together talent technology and capital to drive development The conference sincerely invites experts scholars business people and other relevant personnel from

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