

experiment date(s):

### Experiment 6 - Analysis of an Unknown Chloride

**Goal:** Determine the % chloride (Cl) in an unknown sample.

In the experiment, we will titrate with silver nitrate ( $\text{AgNO}_3$ ) that has a known concentration.

net ionic equation:  $\text{Ag}^+_{(aq)} + \text{Cl}^-_{(aq)} \longrightarrow \text{AgCl}_{(s)}$  white precipitate

When  $\text{Cl}^-$  runs out, the reaction is complete; this is called the **end point**. We will know the end point is reached because  $\text{CrO}_4^{2-}$  (chromate) is added to the chloride solution. It will also form a precipitate with  $\text{Ag}^+$ , but only after all the  $\text{Cl}^-$  is used up

net ionic equation:  $2 \text{Ag}^+_{(aq)} + \text{CrO}_4^{2-}_{(aq)} \longrightarrow \text{Ag}_2\text{CrO}_4_{(s)}$  red precipitate

The chromate ion, yellow in color, will make the titration appear to form yellow precipitate. We will stop adding the silver solution when the chloride solution begins to turn red.

ex. A solid chloride sample weighing 0.3361 g requires 61.46 mL of 0.06258 M  $\text{AgNO}_3$  to reach the end point.

(a) How many mol of  $\text{Cl}^-$  are present in the sample?

$$\frac{0.06146 \text{ L AgNO}_3}{1 \text{ L AgNO}_3} \left( \frac{0.06258 \text{ mol AgNO}_3}{1 \text{ L AgNO}_3} \right) \left( \frac{1 \text{ mol Cl}^-}{1 \text{ mol AgNO}_3} \right) = 3.846 \times 10^{-3} \text{ mol Cl}^-$$

(b) How many g of  $\text{Cl}^-$  are present in the sample?

$$\frac{3.846 \times 10^{-3} \text{ mol Cl}^-}{1 \text{ mol Cl}^-} \left( \frac{35.45 \text{ g Cl}}{1 \text{ mol Cl}} \right) = 0.1363 \text{ g Cl}^-$$

(c) What is the mass % of  $\text{Cl}^-$  in the sample?

$$\frac{\text{g Cl}^-}{\text{g sample}} \times 100 = \frac{0.1363 \text{ g}}{0.3361 \text{ g}} \times 100 = 40.55 \%$$

**Procedure:** omit back titration with NaCl

**Waste disposal:** Dispose of all liquids and rinse all solids into the Heavy Metal Inorganic waste container.

**ASA:** all

# Analysis Of An Unknown Chloride Answers

**AW Rasmussen**



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web the team at thank you letter for hospitality is dedicated to providing the best possible service to our guests and we hope that you will take the time to write a positive review of your stay with us thank you for choosing thank you letter for hospitality

*60 thank you for your hospitality messages that will be* - Apr 11 2023

web here is a list of 60 thank you for your hospitality messages that will be remembered by those who hosted you thank you for a short term visit 1 i would like to sincerely thank you for the amazing company and magnificent welcome into your house it was a pleasure to be over i loved it 2 i enjoyed every single moment you are a marvel

