

# **FA LEVEL 1**

## **IN COACHING FOOTBALL**



# Fa Level 1 Coaching Course

**N Noddings**



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web now realizing that heating and cooling curves represent the amount of heat absorbed or released by a substance during phase changes remember heat uses the variable  $q$  here we have a heating curve versus a cooling curve in a heating curve our substance is absorbing heat

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web nov 15 2021 for the heating and cooling curve of any given substance the solid phase of that substance will be represented in the lower left corner of the graph where the temperature is at its lowest

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to heat 25 00 g of water from 15 0 c to 115 0 c we can determine  $\Delta h$  for the heating process

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web determine the final temperature and state of the ammonia once the system reaches equilibrium assuming that no heat is lost to the surroundings use the information below normal boiling point of  $\text{NH}_3$  33 3 c  $\Delta h_{\text{vap}}$  23 4 kJ/mol specific heat  $\text{NH}_3$  1 4 74 J/g K specific heat  $\text{NH}_3$  2 06 J/g K

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web apr 3 2023 the experiment described above can be summarized in a graph called a heating curve figure below figure 11 16 1 11 16 1 in the heating curve of water the temperature is shown as heat is continually added changes of state occur during plateaus because the temperature is constant

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web the cooling curve a plot of temperature versus cooling time in figure pageindex 4 plots temperature versus time as a 75 g sample of steam initially at 1 atm and 200 c is cooled although we might expect the cooling curve to be the mirror image of the heating curve in figure pageindex 3 the cooling curve is not an

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web once all of the material has solidified at the time indicated by point c the cooling will continue at a rate determined by the heat capacities of the two solids a and b the composition and of course the geometry of the experimental set up by measuring cooling curves for samples of varying composition one can map the entire phase

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web for water this temperature is 100 c because the boiling point for water is 100 c different substances have different melting points and boiling points but the shapes of their heating curves are very similar for example this is the heating curve for iron a metal that melts at 1538 c and boils at 2861 c cooling curves

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web there's 2 on the left and 1 on the right so we need to change the coefficient of  $\text{NH}_3$  to 2 now we have  $\text{N}_2 + \text{H}_2 \rightarrow 2\text{NH}_3$  total the atoms up again on the left there is 2 N and 2 H still on the right there is 2 N and 6 H now so now all we need to do is make the left side have 6 H in total

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web in order to balance a chemical equation the quantities of each type of element and polyatomic ion that are present in the reactants and the products of the reaction must be determined the first reactant in this equation contains magnesium mg and

a hydroxide ion  $\text{OH}^-$  which is a polyatomic anion

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web oct 13 2023 this means that you will need to balance the carbon atoms first 5 use a coefficient to balance the single carbon atom add a coefficient to the single carbon atom on the right of the equation to balance it with the 3 carbon atoms on the left of the equation  $\text{C}_3\text{H}_8 + 2\text{H}_2\text{O} \rightarrow 3\text{CO}_2$

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