

## Unit 1: Atomic Structure

(Also located on \_\_\_\_\_)

D. Ways to describe atomic info:

1. \_\_\_\_\_  
" \_\_\_\_\_

**Example:** An atom has an atomic number of 8. How many protons are present in this atom?

1. \_\_\_\_\_  
" \_\_\_\_\_  
" \_\_\_\_\_

**You Try:**

Atoms of element Hydrogen have: \_\_\_\_\_  
Atoms of element Oxygen have: \_\_\_\_\_  
Atoms of element Iron have: \_\_\_\_\_

1. \_\_\_\_\_  
" \_\_\_\_\_  
" \_\_\_\_\_

**Example:** An atom has 6 protons and 8 neutrons. What is this atom's atomic mass number?

If an atom has a mass number of 20 and an atomic number of 8, how many neutrons are present in this atom?

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**John Allen Berry-Richards**



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**Atomic Structure Teacher's Guide** Tina Youngman,2017      **The Structure of Atoms** Suzanne Slade,2006-08-15  
Explains what an atom is and why it is important and describes the particles that make up atoms      **Elementary Atomic  
Structure** Gordon Kemble Woodgate,1998      *A Guide to Atomic Structure* John Allen Berry-Richards,1972      **Atomic  
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