

VOLUMETRIC ANALYSIS -CLASS XI

EXPERIMENT-1

Aim:

To prepare 100ml $\frac{M}{20}$ sodium carbonate solution and determine the Molarity and the strength of the given solution of hydrochloric acid using $\frac{M}{20}$ sodium carbonate solution.

Apparatus required:

Burette, Pipette, Burette stand, conical flask.

Chemicals Required:

Sodium carbonate, hydrochloric acid, Indicator (methyl orange), distilled water.

Theory:

- To prepare 100ml $\frac{M}{20}$ sodium carbonate solution

Weight of Na_2CO_3 required for preparing 100 ml $M/20$ Na_2CO_3 solution.

$$\frac{M}{20} = \frac{\text{weight of sodium carbonate required for 100 ml} \times 1000}{\text{molar mass of sodium carbonate} \times 100}$$

$$\therefore \frac{1}{20} = \frac{x \times 1000}{106 \times 100} \Rightarrow \frac{x \times 10}{106}$$

$$\frac{1}{20} = \frac{10x}{106}$$

$$= \frac{106}{200} = 0.53 \text{ gm.}$$

Xi Chemistry Practical Procedure Volumetric Analysis

J Elliott



Xi Chemistry Practical Procedure Volumetric Analysis:

Xi Chemistry Practical Procedure Volumetric Analysis Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has be more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Xi Chemistry Practical Procedure Volumetric Analysis**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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