

Chemistry - Chapter 7 Study Guide

Section 7.1:

- *Valence electrons* – electrons in the outer energy level
 - responsible for bonding with other atoms
- How do you know how many valence electrons an element has?
 - Look at the group number
 - Group 1A = 1 valence electron
 - Group 7A = 7 valence electrons
 - Etc...
- *Electron Dot Diagrams* – a way of showing and keeping track of valence electrons
 - Written using the following steps:
 1. Write the element symbol – representing the nucleus and inner electrons
 2. Put one dot for each valence electrons,
 3. Don't forget to add them using Hund's rule – they don't pair up unless they have to.
- *The Octet Rule* – atoms tend to want to achieve a noble gas configuration because it is the most stable. Octet = eight electrons
- *Cations* – a positive ion (usually metals)
 - Lose electrons to attain a noble gas configuration
 - Written like Na^{1+}
 - No dots are shown (on an electron dot diagram) because they lose the valence electrons they had
- *Anions* – a negative ion (usually non-metals)
 - Gain electrons to attain a noble gas configuration
 - Written like P^{3-}
 - Dots are shown because valence electrons are gained
- 8 valence electrons = stable atom (noble gases.)

Section 7.2:

- Anions (-) and cations (+) are held together by opposite charges; a bond is formed through the transfer of electrons to achieve a noble gas configuration (octet.)
- *Ionic Compounds:*
 - Also called salts
 - Made from a cation and an anion
 - Which means they're made of a metal and a non-metal
- ***if you don't understand how to do ionic bonding, check the book or ask somebody to help... it's hard to explain***
- The result of ionic bonding is an ionic compound, shown in a *formula unit*
 - Written like Ca_3P_2

Chemistry Chapter 7 Study Guide For Content Mastery Answers

Yicheng Fang



Chemistry Chapter 7 Study Guide For Content Mastery Answers:

Enjoying the Track of Appearance: An Emotional Symphony within **Chemistry Chapter 7 Study Guide For Content Mastery Answers**

In a world taken by monitors and the ceaseless chatter of instant communication, the melodic beauty and mental symphony developed by the prepared term frequently disappear in to the backdrop, eclipsed by the relentless noise and disruptions that permeate our lives. Nevertheless, set within the pages of **Chemistry Chapter 7 Study Guide For Content Mastery Answers** an enchanting fictional value filled with raw thoughts, lies an immersive symphony waiting to be embraced. Constructed by a wonderful musician of language, that fascinating masterpiece conducts readers on an emotional journey, well unraveling the hidden tunes and profound affect resonating within each cautiously crafted phrase. Within the depths of this poignant evaluation, we can discover the book is central harmonies, analyze their enthralling writing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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