

# Periodic Table Packet #1

Name \_\_\_\_\_

Period \_\_\_\_\_

**Directions:** Use a Periodic table to find the information asked for below:

1. What is the atomic number of:

Calcium \_\_\_\_\_  
Iron \_\_\_\_\_  
Gold \_\_\_\_\_  
Uranium \_\_\_\_\_

2. What is the Atomic mass of:

Calcium \_\_\_\_\_  
Iron \_\_\_\_\_  
Uranium \_\_\_\_\_  
Copper \_\_\_\_\_

3. How many protons do the following have?

Calcium \_\_\_\_\_  
Gold \_\_\_\_\_  
Copper \_\_\_\_\_  
Iron \_\_\_\_\_

4. How many electrons do the following have?

Gold \_\_\_\_\_  
Iron \_\_\_\_\_  
Copper \_\_\_\_\_  
Uranium \_\_\_\_\_

5. Does mercury have more protons and electrons than tin?

6. Is mercury a heavier element than tin?

7. Does potassium have more electrons than neon?

8. Does hydrogen have more electrons than Uranium?

9. Which has more protons, sulfur or iodine?

10. Write the symbols or the names for each of these elements:

Chlorine \_\_\_\_\_

Copper \_\_\_\_\_

Potassium \_\_\_\_\_

Silver \_\_\_\_\_

Na \_\_\_\_\_

Sn \_\_\_\_\_

Zn \_\_\_\_\_

Helium \_\_\_\_\_

Iron \_\_\_\_\_

P \_\_\_\_\_

Ne \_\_\_\_\_

Mercury \_\_\_\_\_

# Periodic Table Packet 1 Answer Key

**Thomas Griffiths**



## Periodic Table Packet 1 Answer Key:

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## **Table of Contents Periodic Table Packet 1 Answer Key**

1. Understanding the eBook Periodic Table Packet 1 Answer Key
  - The Rise of Digital Reading Periodic Table Packet 1 Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Packet 1 Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Packet 1 Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Packet 1 Answer Key
  - Personalized Recommendations
  - Periodic Table Packet 1 Answer Key User Reviews and Ratings
  - Periodic Table Packet 1 Answer Key and Bestseller Lists
5. Accessing Periodic Table Packet 1 Answer Key Free and Paid eBooks

- Periodic Table Packet 1 Answer Key Public Domain eBooks
  - Periodic Table Packet 1 Answer Key eBook Subscription Services
  - Periodic Table Packet 1 Answer Key Budget-Friendly Options
6. Navigating Periodic Table Packet 1 Answer Key eBook Formats
- ePub, PDF, MOBI, and More
  - Periodic Table Packet 1 Answer Key Compatibility with Devices
  - Periodic Table Packet 1 Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Periodic Table Packet 1 Answer Key
  - Highlighting and Note-Taking Periodic Table Packet 1 Answer Key
  - Interactive Elements Periodic Table Packet 1 Answer Key
8. Staying Engaged with Periodic Table Packet 1 Answer Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Packet 1 Answer Key
9. Balancing eBooks and Physical Books Periodic Table Packet 1 Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Packet 1 Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Packet 1 Answer Key
- Setting Reading Goals Periodic Table Packet 1 Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Packet 1 Answer Key
- Fact-Checking eBook Content of Periodic Table Packet 1 Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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