

**TEACHING TRANSPARENCY WORKSHEET**

**9**

# Types of Matter

**Use with Chapter 3,  
Section 3.4**

**1.** Into what two broad classes can all matter be divided?  
\_\_\_\_\_

**2.** What is the difference between a mixture and a pure substance?  
\_\_\_\_\_

**3.** What is the difference between a compound and an element?  
\_\_\_\_\_

**4.** Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.  
\_\_\_\_\_

**5.** A list of compounds and elements is given below. Circle the substances that are elements.

- |       |          |          |
|-------|----------|----------|
| gold  | aluminum | water    |
| sugar | oxygen   | platinum |
| salt  | chlorine | brass    |

**6.** How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?  
\_\_\_\_\_

**7.** Label each mixture below as either homogeneous or heterogeneous.

- |  |  |
|--|--|
| <p><b>a.</b> air _____</p> <p><b>b.</b> clay _____</p> <p><b>c.</b> homemade lemonade (with pulp) _____</p> <p><b>d.</b> oatmeal raisin cookie _____</p> | <p><b>e.</b> finger paint _____</p> <p><b>f.</b> vinegar _____</p> <p><b>g.</b> soil _____</p> |
|--|--|

**8.** List three methods that are commonly used to separate mixtures into their component substances.  
\_\_\_\_\_

# Teaching Transparency Master 2 Answer Key

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