

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Matching

Match each item with the correct statement below.

Match each item with the correct statement below.

- | | |
|--------------------------------|--------------------|
| a. monatomic ion | f. cation |
| b. acid | g. binary compound |
| c. base | h. anion |
| d. law of definite proportions | i. polyatomic ion |
| e. law of multiple proportions | |

- ___ 1. consists of a single atom with a positive or negative charge
- ___ 2. atom or group of atoms having a negative charge
- ___ 3. atom or group of atoms having a positive charge
- ___ 4. tightly-bound group of atoms that behaves as a unit and carries a net charge
- ___ 5. compound composed of two different elements
- ___ 6. produces a hydrogen ion when dissolved in water
- ___ 7. produces a hydroxide ion when dissolved in water
- ___ 8. In any chemical compound, the masses of elements are always in the same proportion by mass.
- ___ 9. when two elements form more than one compound, the masses of one element that combine with the same mass of the other element are in the ratio of small, whole numbers

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ___ 10. When naming a transition metal ion that can have more than one common ionic charge, the numerical value of the charge is indicated by a ____.
- | | |
|-----------|-------------------------------------|
| a. prefix | c. Roman numeral following the name |
| b. suffix | d. superscript after the name |
- ___ 11. Which of the following correctly provides the name of the element, the symbol for the ion, and the name of the ion?
- | | |
|-----------------------------------|-------------------------------------|
| a. fluorine, F^+ , fluoride ion | c. copper, Cu^+ , cuprous ion |
| b. zinc, Zn^{2+} , zincate ion | d. sulfur, S^{2-} , sulfurous ion |
- ___ 12. The nonmetals in Groups 6A and 7A ____.
- | |
|---|
| a. lose electrons when they form ions |
| b. have a numerical charge that is found by subtracting 8 from the group number |
| c. all have ions with a -1 charge |
| d. end in <i>-ite</i> |
- ___ 13. Which of the following is NOT a cation?
- | | |
|------------------|------------------|
| a. iron(III) ion | c. Ca^{2+} |
| b. sulfate | d. mercurous ion |

Chapter 9 Practice Test Naming And Writing Chemical Formulas

Peter P. Berlow



Chapter 9 Practice Test Naming And Writing Chemical Formulas:

Decoding **Chapter 9 Practice Test Naming And Writing Chemical Formulas**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chapter 9 Practice Test Naming And Writing Chemical Formulas**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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