

## Section 1.1

### Section 1.1

Section 1.1 covers the following questions in the space provided:

1. ... 1. ... The following are chemical equations representing the reaction of hydrogen gas with oxygen gas:
  - (a)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (b)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (c)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (d)  $2H_2 + O_2 \rightarrow 2H_2O$
2. ... 2. ... The following are chemical equations representing the reaction of hydrogen gas with oxygen gas:
  - (a)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (b)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (c)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (d)  $2H_2 + O_2 \rightarrow 2H_2O$
3. ... 3. ... The following are chemical equations representing the reaction of hydrogen gas with oxygen gas:
  - (a)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (b)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (c)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (d)  $2H_2 + O_2 \rightarrow 2H_2O$
4. ... 4. ... The following are chemical equations representing the reaction of hydrogen gas with oxygen gas:
  - (a)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (b)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (c)  $2H_2 + O_2 \rightarrow 2H_2O$
  - (d)  $2H_2 + O_2 \rightarrow 2H_2O$

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$$2H_2 + O_2 \rightarrow 2H_2O$$

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# Chapter 9 Section 1 Review Stoichiometry Answers

**Thirumalaisamy P.  
Velavancorresponding**



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