

WORKSHEET
CHEMISTRY FOR ENGINEERING STUDENTS, LECTURE 1
 CHEMISTRY OF METALS AND ALLOYS

ANSWERS KEY

QUESTION 1. CLASSIFY THE ELEMENTS OF PERIODIC TABLE INTO METALS AND NON-METALS.

1. Distinguish between metals and non-metals based on properties such as electrical and thermal conductivity, ductility, malleability, etc.
 - (a) Write the balanced chemical equation for the reaction of Cu with H_2SO_4 and HNO_3 .
 - (i) $\text{Cu} + \text{H}_2\text{SO}_4 \rightarrow \text{CuSO}_4 + \text{H}_2$
 - (ii) $3\text{Cu} + 8\text{HNO}_3 \rightarrow 3\text{Cu(NO}_3)_2 + 2\text{N}_2\text{O} + 4\text{H}_2\text{O}$
 - (b) Write the balanced chemical equation for the reaction of Fe with H_2SO_4 and HNO_3 .
 - (i) $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{FeSO}_4 + \text{H}_2$
 - (ii) $\text{Fe} + 4\text{HNO}_3 \rightarrow \text{Fe(NO}_3)_3 + \text{NO} + 2\text{H}_2\text{O}$
2. State the main differences between ionic and covalent compounds.
 - (a) Write the balanced chemical equation for the reaction of NaCl with H_2O .
 - (i) $\text{NaCl} + \text{H}_2\text{O} \rightarrow \text{Na}^+ + \text{Cl}^-$
 - (b) Write the balanced chemical equation for the reaction of H_2O with CO_2 .
 - (i) $\text{H}_2\text{O} + \text{CO}_2 \rightarrow \text{H}_2\text{CO}_3$
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