

## Assessment

## Chapter Test A

## Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
  - a slippery feel
  - the ability to conduct an electric current
  - reactivity with metals
- \_\_\_\_\_ 2. Which is an example of a weak acid?
- HCl
  - H<sub>2</sub>SO<sub>4</sub>
  - H<sub>3</sub>PO<sub>4</sub>
  - HNO<sub>3</sub>
- \_\_\_\_\_ 3. Acetic acid is found in
- vinegar.
  - the stomach.
  - antacids.
  - oranges.
- \_\_\_\_\_ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
  - tastes bitter.
  - dissociates to form OH<sup>-</sup> ions.
  - All of the above
- \_\_\_\_\_ 5. Which of the following is a Brønsted-Lowry base?
- HCl
  - HCO<sub>3</sub><sup>-</sup>
  - H<sub>3</sub>O<sup>+</sup>
  - H<sub>3</sub>PO<sub>4</sub>
- \_\_\_\_\_ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4^-(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
- which of these is a conjugate acid-base pair?
- H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>(aq) and CH<sub>3</sub>NH<sub>2</sub>(aq)
  - H<sub>2</sub>C<sub>2</sub>O<sub>4</sub>(aq) and CH<sub>3</sub>NH<sub>3</sub><sup>+</sup>(aq)
  - CH<sub>3</sub>NH<sub>2</sub>(aq) and CH<sub>3</sub>NH<sub>3</sub><sup>+</sup>(aq)
  - HC<sub>2</sub>O<sub>4</sub><sup>-</sup>(aq) and CH<sub>3</sub>NH<sub>3</sub><sup>+</sup>(aq)

# Modern Chemistry Chapter 7 Test Answer Key

**M Carnoy**



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