

Mixture Problem – Remove from Solution

John has 20 ounces of a 20% of salt solution.

How much water should be evaporated to make it a 30% solution?

20% of salt solution = 80% of water.

30% of salt solution = 70% of water.

Let y be the amount of water evaporated.

	Original	Removed	Result
Water	80% = 0.8	100% = 1	70% = 0.70
Amount	20	y	$20 - y$
Multiply	0.8×20	$1 \times y$	$0.70(20 - y)$

Original - Removed = Result

$$(0.8 \times 20) - (1 \times y) = 0.70(20 - y)$$

$$16 - y = 14 - 0.7y$$

$$y - 0.7y = 16 - 14$$

$$0.3y = 2$$

$$y = \frac{2}{0.3} \approx 6.67$$

Answer: 6.67 ounces of water should be evaporated.

Mixture Problems With Solutions And Answers

Scott C. Dulebohn

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular glow behind it.

Mixture Problems With Solutions And Answers:

Pharmaceutical Calculations Michalakis Savva, 2019-10-17 Pharmaceutical Calculations A Conceptual Approach is a book that combines conceptual and procedural understanding for students and will guide you to master prerequisite skills to carry out accurate compounding and dosage regimen calculations It is a book that makes the connection between basic sciences and pharmacy It describes the most important concepts in pharmaceutical sciences thoroughly accurately and consistently through various commentaries and activities to make you a scientific thinker and to help you succeed in college and licensure exams Calculation of the error associated with a dose measurement can only be carried out after understanding the concept of accuracy versus precision in a measurement Similarly full appreciation of drug absorption and distribution to tissues can only come about after understanding the process of transmembrane passive diffusion Early understanding of these concepts will allow reinforcement and deeper comprehension of other related concepts taught in other courses More weight is placed on the qualitative understanding of fundamental concepts like tonicity vs osmotic pressure diffusion vs osmosis crystalloids vs colloids osmotic diuretics vs plasma expanders rate of change vs rate constants drug accumulation vs drug fluctuation loading dose vs maintenance dose body surface area BSA vs body weight BW as methods to adjust dosages and much more before considering other quantitative problems In one more significant innovation the origin and physical significance of all final forms of critical equations is always described in detail thus allowing recognition of the real application and limitations of an equation Specific strategies are explained step by step in more than 100 practice examples taken from the fields of compounding pharmacy pharmaceuticals pharmacokinetics pharmacology and medicine

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