

PRE-LAB QUESTIONS

Hydrates

1. If a hydrate of the formula $MCl_x \cdot xH_2O$ decomposes when heated to produce HCl, what change would you expect to occur when a piece of blue litmus paper is held in the path of the vapor that is released? Explain.

2. After 0.6523 g of $CoCl_2 \cdot 6H_2O$ is heated, the residue has a mass of 0.3423 g. Calculate the percent H_2O in the hydrate. What is the formula of the hydrate?

3. What safety precautions are necessary in this experiment?

4. $CoCl_2$ is often used in hygrometers. Search the Internet to determine why. How does this relate to the experiment?

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Christian G. Meyer



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