

**TEACHING TRANSPARENCY WORKSHEET**

**9**

# Types of Matter

**Use with Chapter 3,  
Section 3.4**

**1.** Into what two broad classes can all matter be divided?  
\_\_\_\_\_

**2.** What is the difference between a mixture and a pure substance?  
\_\_\_\_\_

**3.** What is the difference between a compound and an element?  
\_\_\_\_\_

**4.** Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.  
\_\_\_\_\_

**5.** A list of compounds and elements is given below. Circle the substances that are elements.

- |       |          |          |
|-------|----------|----------|
| gold  | aluminum | water    |
| sugar | oxygen   | platinum |
| salt  | chlorine | brass    |

**6.** How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?  
\_\_\_\_\_

**7.** Label each mixture below as either homogeneous or heterogeneous.

- |                                               |                              |
|-----------------------------------------------|------------------------------|
| <b>a.</b> air _____                           | <b>e.</b> finger paint _____ |
| <b>b.</b> clay _____                          | <b>f.</b> vinegar _____      |
| <b>c.</b> homemade lemonade (with pulp) _____ | <b>g.</b> soil _____         |
| <b>d.</b> oatmeal raisin cookie _____         |                              |

**8.** List three methods that are commonly used to separate mixtures into their component substances.  
\_\_\_\_\_

# Teaching Transparency Chemistry Chapter 19

**SA Adler**



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## **Table of Contents Teaching Transparency Chemistry Chapter 19**

1. Understanding the eBook Teaching Transparency Chemistry Chapter 19
  - The Rise of Digital Reading Teaching Transparency Chemistry Chapter 19
  - Advantages of eBooks Over Traditional Books
2. Identifying Teaching Transparency Chemistry Chapter 19
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Teaching Transparency Chemistry Chapter 19
  - User-Friendly Interface

4. Exploring eBook Recommendations from Teaching Transparency Chemistry Chapter 19
  - Personalized Recommendations
  - Teaching Transparency Chemistry Chapter 19 User Reviews and Ratings
  - Teaching Transparency Chemistry Chapter 19 and Bestseller Lists
5. Accessing Teaching Transparency Chemistry Chapter 19 Free and Paid eBooks
  - Teaching Transparency Chemistry Chapter 19 Public Domain eBooks
  - Teaching Transparency Chemistry Chapter 19 eBook Subscription Services
  - Teaching Transparency Chemistry Chapter 19 Budget-Friendly Options
6. Navigating Teaching Transparency Chemistry Chapter 19 eBook Formats
  - ePub, PDF, MOBI, and More
  - Teaching Transparency Chemistry Chapter 19 Compatibility with Devices
  - Teaching Transparency Chemistry Chapter 19 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Teaching Transparency Chemistry Chapter 19
  - Highlighting and Note-Taking Teaching Transparency Chemistry Chapter 19
  - Interactive Elements Teaching Transparency Chemistry Chapter 19
8. Staying Engaged with Teaching Transparency Chemistry Chapter 19
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Teaching Transparency Chemistry Chapter 19
9. Balancing eBooks and Physical Books Teaching Transparency Chemistry Chapter 19
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Teaching Transparency Chemistry Chapter 19
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Teaching Transparency Chemistry Chapter 19
  - Setting Reading Goals Teaching Transparency Chemistry Chapter 19
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Teaching Transparency Chemistry Chapter 19
  - Fact-Checking eBook Content of Teaching Transparency Chemistry Chapter 19
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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