

## Section 12.1

### Section 12.1

Answer complete answer multiple-choice questions that are provided.

1. \_\_\_\_\_
  - a. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
  - b. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
2. \_\_\_\_\_
  - a. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
  - b. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
3. \_\_\_\_\_
  - a. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_
  - b. \_\_\_\_\_
    - (1) \_\_\_\_\_
    - (2) \_\_\_\_\_
    - (3) \_\_\_\_\_
    - (4) \_\_\_\_\_

4. Given the two reactions and the following balanced equation:



a. Balance the equation:



b. What is the coefficient of  $\text{H}_2$  in  $\text{H}_2$ ?

c. How many molecules of  $\text{H}_2$  are produced from 100 molecules of  $\text{CO}$  and 100 molecules of  $\text{H}_2\text{O}$ ?

d. How many molecules of  $\text{CO}_2$  are produced from 100 molecules of  $\text{H}_2$  and 100 molecules of  $\text{CO}$ ?

e. How many grams of  $\text{H}_2$  are produced from 100 grams of  $\text{H}_2\text{O}$  and 100 grams of  $\text{CO}$ ?

# Chapter 9 Stoichiometry Section 1 Answers

**JR Anderson**



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