

## Thermodynamic Equilibrium - 1 (Application of Equilibrium)

### Thermodynamic Equilibrium

Thermodynamic Equilibrium states that both conditions are necessary to produce the equilibrium:

- 1) An absence of net work and energy production (thermodynamically) and/or the process is reversible.  
 **Equal**  
 None  
 All of the above
- 2) The two thermodynamically stable states are at the same pressure and at the same temperature.  
 **Equal**  
 None  
 All of the above
- 3) There is no net work production and/or the two states are at the same pressure and temperature.  
 None  
 Both  
 **Both**  
 None of the above
- 4) There is no net work production and/or the two states are at the same pressure and temperature.  
 None  
 Both  
 **Both**  
 None of the above
- 5) There is no net work production and/or the two states are at the same pressure and temperature.  
 Equal temperature of states  
 Equal and homogeneous  
 Equal force  
 **Equal temperature of homogeneous**
- 6) There is no net work production and/or the two states are at the same pressure and temperature.  
 **Equal**  
 None  
 Both  
 All of the above
- 7) There is no net work production and/or the two states are at the same pressure and temperature.  
 None  
 Both  
 **Both**  
 None of the above
- 8) There is no net work production and/or the two states are at the same pressure and temperature.  
 Equal temperature of states  
 Equal and homogeneous  
 Equal force  
 **Equal temperature of homogeneous**

# Chapter Test Thermal Energy Answers

**Jianjun Gao**



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