

**UNIT 10: PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE** **PERIODIC TABLE**

**Section 10.1** (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____
_____

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

**Column A**

**Column B**

- |   |                                  |
|---|----------------------------------|
| _____ 18. A column in the periodic table                  | _____ a. metals                  |
| _____ 19. A row in the periodic table                     | _____ b. group                   |
| _____ 20. Group 1 elements                                | _____ c. period                  |
| _____ 21. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 22. Group 8 elements                                | _____ e. transition elements     |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of platinum.
- \_\_\_\_\_ 24. Most nonmetals form diatomic molecules in the periodic table (or naturally).
- \_\_\_\_\_ 25. Group 1A elements except for hydrogen are located on the left side.
- \_\_\_\_\_ 26. Group 2A elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 7A elements are highly reactive nonmetals known as halogens.
- \_\_\_\_\_ 28. Group 8A elements are inert nonmetallic elements known as noble gases.
- \_\_\_\_\_ 29. Alkaloids have properties of both metals and more reactive metals.

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**Rachel S Tattersall**



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### **Table of Contents Chemistry Chapter 6 Study Guide For Content Mastery Answers**

1. Understanding the eBook Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - The Rise of Digital Reading Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Personalized Recommendations
  - Chemistry Chapter 6 Study Guide For Content Mastery Answers User Reviews and Ratings
  - Chemistry Chapter 6 Study Guide For Content Mastery Answers and Bestseller Lists
5. Accessing Chemistry Chapter 6 Study Guide For Content Mastery Answers Free and Paid eBooks
  - Chemistry Chapter 6 Study Guide For Content Mastery Answers Public Domain eBooks
  - Chemistry Chapter 6 Study Guide For Content Mastery Answers eBook Subscription Services

- Chemistry Chapter 6 Study Guide For Content Mastery Answers Budget-Friendly Options
- 6. Navigating Chemistry Chapter 6 Study Guide For Content Mastery Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Chapter 6 Study Guide For Content Mastery Answers Compatibility with Devices
  - Chemistry Chapter 6 Study Guide For Content Mastery Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Highlighting and Note-Taking Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Interactive Elements Chemistry Chapter 6 Study Guide For Content Mastery Answers
- 8. Staying Engaged with Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Chapter 6 Study Guide For Content Mastery Answers
- 9. Balancing eBooks and Physical Books Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Chapter 6 Study Guide For Content Mastery Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Setting Reading Goals Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Fact-Checking eBook Content of Chemistry Chapter 6 Study Guide For Content Mastery Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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