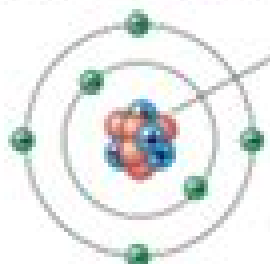


## Atomic structure

Label the following diagram with the **correct** proton, neutron, electron and nucleus.



Carbon atom

2. Now complete the table below, identifying the charges, mass and place found for each of the sub-atomic particles.

Sub-atomic particle	Place found	Mass	Charge
proton			
neutron			
electron			

3. Why are atoms neutral overall?

.....

.....

## Atomic structure 2

Use the information in the periodic table to help you complete the table below:

Element	Protons	Neutrons	Electrons
Carbon			
Sodium			
Magnesium			
<b>Isotope</b>			
Bromine			
Hydrogen			
Aluminium			
Argon			

3. What is an isotope?

.....

.....

## Calculating Relative Atomic Mass "A<sub>r</sub>"

Calculate the relative atomic mass of the following elements, given their mass and abundance.

a) Bromine, consisting of 50% Br-79 and 50% Br-81

.....

b) Copper, consisting of 69.0% Cu-63 and 30.9% Cu-65

.....

c) Silver, consisting of 97.5% Ag-107 and 2.5% Ag-109

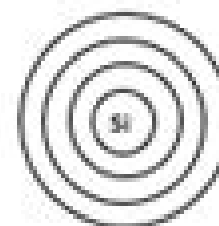
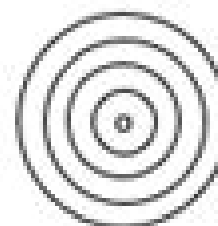
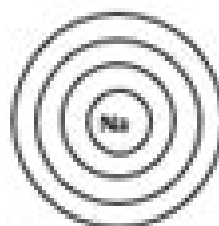
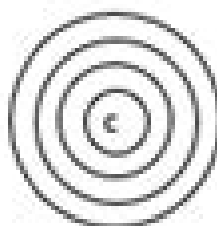
.....

## Electron structure

1. How many electrons can each shell hold?

a) 1<sup>st</sup> shell ..... b) 2<sup>nd</sup> shell ..... c) 3<sup>rd</sup> shell ..... d) 4<sup>th</sup> shell .....

2. Complete the electron structure diagrams for the following elements by drawing electrons on the shells. Under each diagram, WRITE the electron configuration, e.g. 2, 8, 1



## The Periodic Table

1. Shade the metals in blue

2. Shade the non-metals in red

3. Shade the metalloids in green

4. What is special about the properties of metalloids?

.....

## The Periodic Table

1. What element is found in:

a) Group 2 Period 4 .....

b) Group 1 Period 1 .....

c) Group 7 Period 2 .....

d) Group 8 Period 3 .....

2. What group and period are the following elements found in?

a) Lithium .....

b) Phosphorus .....

c) Iron .....

d) Rubidium .....

# Chemistry End Off Topic Assessment Unit C2

**Yan Bai**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle positioned behind the bar's end.

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