

Name:

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## Chapter 17 & 19 Study Guide

### Key Terms

*Directions: Define each of the following key terms from chapters seventeen and nineteen on a separate sheet of paper.*

- **Atom** - the smallest piece of matter that still has the properties of the element
- **atomic number** – number of protons in an atom's nucleus
- **atomic mass** – average mass of all an element's isotopes
- **ductile** – capable of being drawn into wire
- **electron** – subatomic particle with a negative charge
- **electron cloud** – area outside the nucleus in which electrons can be found.
- **Electron dot diagram** – uses the element's symbol and dots to represent the valence electrons in an element.
- **energy level** – areas within the electron cloud where electrons are most likely to be found.
- **Family** – vertical columns of the periodic table
- **Group** - vertical columns of the periodic table
- **half-life** – the amount of time it takes half the sample of a radioactive isotope to decay.
- **Isotope** – atoms of the same element that have different numbers of neutrons.
- **Lustrous** – having a shiny appearance
- **Malleable** – can be hammered flat or rolled into sheets
- **mass number** – sum of protons and neutrons in the nucleus of an atom.
- **Metal** – element that typically is hard, shiny, solid, malleable, and is a good conductor of heat and electricity.
- **Neutron** – neutrally charged subatomic particle.
- **Nonmetal** - element that typically is brittle solid or gas, not shiny, not malleable, and is a poor conductor of heat and electricity.
- **Nucleus** – center of the atom
- **Period** – horizontal row of the periodic table.
- **periodic table** – organized list of all known elements.
- **Proton** – positively charged subatomic particle
- **transition metal** – Elements found in families 3-12 of the periodic table.
- **transuranium element** – an element containing more than 92 protons in its nucleus
- **quark** – particle believed to make up protons and neutrons.
- **valence electron** – electrons found in the outermost energy level of an atom.

# Chapter 17 Study Guide Answers

**B Lingard**



## **Chapter 17 Study Guide Answers:**

## Whispering the Secrets of Language: An Mental Quest through **Chapter 17 Study Guide Answers**

In a digitally-driven earth where screens reign supreme and quick transmission drowns out the subtleties of language, the profound secrets and psychological subtleties hidden within phrases often go unheard. However, located within the pages of **Chapter 17 Study Guide Answers** a fascinating fictional value sporting with fresh feelings, lies a fantastic quest waiting to be undertaken. Written by a talented wordsmith, this charming opus attracts visitors on an introspective journey, gently unraveling the veiled truths and profound impact resonating within ab muscles fabric of each word. Within the mental depths of this touching review, we will embark upon a honest exploration of the book is core styles, dissect its interesting publishing fashion, and succumb to the strong resonance it evokes serious within the recesses of readers hearts.

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