

Name \_\_\_\_\_

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 1) Formation of solutions where the process is endothermic can be spontaneous provided that \_\_\_\_\_.
  - A) the solvent is a gas and the solute is a solid
  - B) they are accompanied by an increase in order
  - C) they are accompanied by an increase in disorder
  - D) they are accompanied by another process that is exothermic
  - E) the solvent is water and the solute is a gas
- 2) The phrase "like dissolves like" refers to the fact that \_\_\_\_\_.
  - A) polar solvents dissolve nonpolar solutes and vice versa
  - B) solvents can only dissolve solutes of similar molar mass
  - C) gases can only dissolve other gases
  - D) polar solvents dissolve polar solutes and nonpolar solvents dissolve nonpolar solutes
  - E) condensed phases can only dissolve other condensed phases
- 3) An unsaturated solution is one that \_\_\_\_\_.
  - A) has no double bonds
  - B) has a concentration lower than the solubility
  - C) contains no solute
  - D) contains the maximum concentration of solute possible, and is in equilibrium with undissolved solute
  - E) contains more dissolved solute than the solubility allows
- 4) A solution with a concentration higher than the solubility is \_\_\_\_\_.
  - A) is not possible
  - B) is saturated
  - C) is supercritical
  - D) is unsaturated
  - E) is supersaturated
- 5) Which one of the following substances would be the most soluble in  $\text{CCl}_4$ ?  
A)  $\text{H}_2\text{O}$                       B)  $\text{NaCl}$                       C)  $\text{CH}_3\text{CH}_2\text{OH}$                       D)  $\text{NH}_3$                       E)  $\text{C}_{10}\text{H}_{22}$
- 6) Which one of the following is most soluble in water?
  - A)  $\text{CH}_3\text{OH}$
  - B)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$
  - C)  $\text{CH}_3\text{CH}_2\text{OH}$
  - D)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$
  - E)  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$

# Chapter 11 Chemistry Test

**ML Yell**



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