

9. (b) Bases soluble in water are called alkalis. Only the oxides of sodium, potassium, and calcium are soluble in water, so these form sodium hydroxide potassium hydroxide and calcium hydroxide. These are the strongest bases.

10. **Acids react with metals to liberate**
gas

- (a) Oxygen (b) Carbon dioxide
(c) Hydrogen (d) Nitrogen

10. (c) All acids react with metals to liberate hydrogen gas
Metal + Acid → Metal salt + Hydrogen gas

11. **Choose the correct statement(s)**

- (i) Most of the acids are water soluble
(ii) Acids react with metallic oxides and hydroxides to form metallic salt and water only.
(iii) Acids react with metallic carbonates to form metallic salt and hydrogen gas and water
(iv) Acetic acid is used as a food preservative
(a) (i) & (ii) only (b) (iii) & (iv)
(c) (i), (ii) & (iv) (d) all the above

11. (c) The third statement is wrong because acid reacts with metal carbonates to form metallic salt, carbon dioxide gas and water

12. **The reaction of acid and a base to form salt and water only is called**

- (a) Neutralization (b) Addition
(c) Substitution (d) Oxidation

12. (a) A neutralization reaction is the one in which acid and base react to form neutral salt and water.

13. **When ammonia gas is dissolved in water it reacts to form**

- (a) Ammonia water
(b) Ammonium hydroxide
(c) Ammonia acid
(d) All the above

13. (b) Ammonia reacts with water to form a hydroxide called ammonium hydroxide (NH₄OH)



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Table of Contents Chapter 14 Acids And Bases Worksheet

1. Understanding the eBook Chapter 14 Acids And Bases Worksheet
 - The Rise of Digital Reading Chapter 14 Acids And Bases Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 14 Acids And Bases Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 14 Acids And Bases Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 14 Acids And Bases Worksheet
 - Personalized Recommendations
 - Chapter 14 Acids And Bases Worksheet User Reviews and Ratings
 - Chapter 14 Acids And Bases Worksheet and Bestseller Lists

5. Accessing Chapter 14 Acids And Bases Worksheet Free and Paid eBooks
 - Chapter 14 Acids And Bases Worksheet Public Domain eBooks
 - Chapter 14 Acids And Bases Worksheet eBook Subscription Services
 - Chapter 14 Acids And Bases Worksheet Budget-Friendly Options
6. Navigating Chapter 14 Acids And Bases Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 14 Acids And Bases Worksheet Compatibility with Devices
 - Chapter 14 Acids And Bases Worksheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 14 Acids And Bases Worksheet
 - Highlighting and Note-Taking Chapter 14 Acids And Bases Worksheet
 - Interactive Elements Chapter 14 Acids And Bases Worksheet
8. Staying Engaged with Chapter 14 Acids And Bases Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 14 Acids And Bases Worksheet
9. Balancing eBooks and Physical Books Chapter 14 Acids And Bases Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 14 Acids And Bases Worksheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 14 Acids And Bases Worksheet
 - Setting Reading Goals Chapter 14 Acids And Bases Worksheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 14 Acids And Bases Worksheet
 - Fact-Checking eBook Content of Chapter 14 Acids And Bases Worksheet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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