

Reaction Energy

Section 1

Answer **ALL** of the following questions in the space provided.

- The following reaction is exothermic. What is the value of ΔH_{rxn} ?
- The following reaction is exothermic. What is the value of ΔH_{rxn} ?



- Is energy being released or is energy being absorbed? Liquid H_2O at room temperature?
- Is the reaction exothermic or endothermic? Is the reaction spontaneous or non-spontaneous at room temperature?

Answer **ALL** of the following questions in the box to the left. Answer all your work in this space provided.

- 10.000 g of water at 20.0°C absorbs a certain amount of energy when it is heated to 100.0°C .

- What is the specific heat capacity of water (e.g. H_2O)? Does water absorb energy or release energy when heated? Write the chemical formula of water and show the change from 20.0°C to 100.0°C .

- The following is a decomposition reaction. What is the value of ΔH_{rxn} ?



Chapter 16 Review Reaction Energy Section 1 Short Answer

Shasha Hu



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