

CHAPTER 16 REVIEW

Chemical Equilibrium

Teacher Notes and Answers

Chapter 16 Introduction 1

Question 1

- a. $\frac{[CO_2]}{[CO]}$
- b. $\frac{[CO_2]^2}{[CO]}$

- c. $\frac{[CO_2]^2}{[CO]}$
- d. $\frac{[CO_2]^2}{[CO]}$

e. $\frac{[CO_2]^2}{[CO]}$ (wrong)

f. $\frac{[CO_2]^2}{[CO]}$

g. $\frac{[CO_2]^2}{[CO]}$ (wrong)

- h. correct
- i. correct

- 11. a. The value of the K_c expression is constant. The value of the K_c expression is 1.0. The K_c expression is the product of the concentrations.

- b. 1.0
- c. 1.0

Introduction 2

Question 1

- a. $\frac{[CO_2]}{[CO]}$
- b. $\frac{[CO_2]^2}{[CO]}$
- c. $\frac{[CO_2]^2}{[CO]}$

- d. correct
- e. correct
- f. correct
- g. correct

- 11. a. These values are not constant concentrations. Concentrations are constant. The value of the K_c expression is constant. The value of the K_c expression is 1.0.

- b. The value of the K_c expression is constant. The value of the K_c expression is 1.0. The value of the K_c expression is the product of the concentrations.

- c. The value of the K_c expression is constant. The value of the K_c expression is 1.0. The value of the K_c expression is the product of the concentrations.

- d. The value of the K_c expression is constant. The value of the K_c expression is 1.0. The value of the K_c expression is the product of the concentrations.

- e. The value of the K_c expression is constant. The value of the K_c expression is 1.0. The value of the K_c expression is the product of the concentrations.

- f. The value of the K_c expression is constant. The value of the K_c expression is 1.0. The value of the K_c expression is the product of the concentrations.

Question 2

- a. $\frac{[CO_2]}{[CO]}$
- b. $\frac{[CO_2]^2}{[CO]}$
- c. $\frac{[CO_2]^2}{[CO]}$
- d. $\frac{[CO_2]^2}{[CO]}$

- e. $\frac{[CO_2]^2}{[CO]}$ (wrong)
- f. $\frac{[CO_2]^2}{[CO]}$ (wrong)

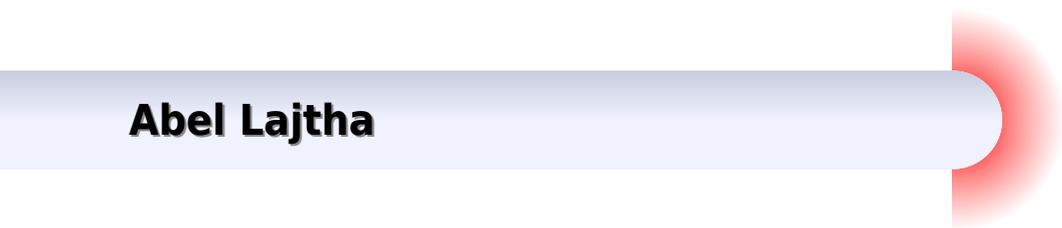
- g. $\frac{[CO_2]^2}{[CO]}$
- h. $\frac{[CO_2]^2}{[CO]}$

- i. $\frac{[CO_2]^2}{[CO]}$ (wrong)
- j. $\frac{[CO_2]^2}{[CO]}$ (wrong)

- k. $\frac{[CO_2]^2}{[CO]}$
- l. $\frac{[CO_2]^2}{[CO]}$

Chapter 18 Review Chemical Equilibrium Section 3 Answers

Abel Lajtha



Chapter 18 Review Chemical Equilibrium Section 3 Answers:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Alterations of Chemical Equilibrium in the Nervous System** Abel Lajtha, 2013-11-21 It has been recognized for more than a thousand years that the function of the brain like the function of the other organs of the body is determined by its physical chemical and biological properties Evidence that even its highest functions could be explained by these properties was gathered only in recent years however these findings which clearly have to be confirmed by a great deal of further experimental evidence indicate that most if not all of the functions of the brain are based on its bio chemical and biophysical mechanisms This at first hearing may sound rather simple but the ability to understand learning emotion perhaps even creativity on biological terms may well be the most important scientific discovery of all time Few pieces of knowledge can influence our future health and well being to the degree that understanding of mental mechanisms will It has been clearly shown in many ways in the previous volumes of this Handbook that from the biochemical or neurochemical point of view the brain is one of the most active organs The brain seems stable and in some respects permanent this is evidence not of inactivity but of carefully controlled homeostasis of dynamic rather than static equilibrium with most components undergoing metabolic alterations

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