

ITIL Practice Exam With Questions and Answers

What are the objectives of service level management? - ANSWER 1. Defining, documenting and agreeing the level of IT services to be provided.

2. Monitoring, measuring and reporting the actual level of services provided.

3. Monitoring and improving customer satisfaction.

What is the result of carrying out an activity, following a process or delivering an IT known as? - ANSWER Outcome

A process owner has been identified with an "I" in a RACI matrix. Which of the following is expected of them? - ANSWER Responsible, Accountable, Consulted, Informed.

Be kept up-to-date on the progress of an activity.

Which one of the following do technology metrics measure? - ANSWER Components

Which process includes business, service and component sub-processes? - ANSWER Capacity management

Which one of the following is NOT part of the service design stage of the service lifecycle? - ANSWER Designing and maintaining all necessary service transition packages.

Which process is responsible for managing relationships with vendors? - ANSWER Supplier management

Which of the following service desk organizational structures are described in service operation? - ANSWER Local service desk, virtual service desk, follow the sun

What are the categories of event described in the ITIL service operation book? - ANSWER Informational, warning, exception

What would you call the groups of people who have an interest in the activities, targets, resources and deliverables from service management? - ANSWER

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**Thirumalaisamy P.
Velavancorresponding**

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web donor whereas a base generically rep resented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids e g sulfuric acid have a very pronounced ten dency to transfer a proton to a base whereas weak acids

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral ph at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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