

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Matching

Match each item with the correct statement below.

Match each item with the correct statement below.

- | | |
|--------------------------------|--------------------|
| a. monatomic ion | f. cation |
| b. acid | g. binary compound |
| c. base | h. anion |
| d. law of definite proportions | i. polyatomic ion |
| e. law of multiple proportions | |

- ___ 1. consists of a single atom with a positive or negative charge
- ___ 2. atom or group of atoms having a negative charge
- ___ 3. atom or group of atoms having a positive charge
- ___ 4. tightly-bound group of atoms that behaves as a unit and carries a net charge
- ___ 5. compound composed of two different elements
- ___ 6. produces a hydrogen ion when dissolved in water
- ___ 7. produces a hydroxide ion when dissolved in water
- ___ 8. In any chemical compound, the masses of elements are always in the same proportion by mass.
- ___ 9. when two elements form more than one compound, the masses of one element that combine with the same mass of the other element are in the ratio of small, whole numbers

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ___ 10. When naming a transition metal ion that can have more than one common ionic charge, the numerical value of the charge is indicated by a ____.
- | | |
|-----------|-------------------------------------|
| a. prefix | c. Roman numeral following the name |
| b. suffix | d. superscript after the name |
- ___ 11. Which of the following correctly provides the name of the element, the symbol for the ion, and the name of the ion?
- | | |
|-----------------------------------|-------------------------------------|
| a. fluorine, F^+ , fluoride ion | c. copper, Cu^+ , cuprous ion |
| b. zinc, Zn^{2+} , zincate ion | d. sulfur, S^{2-} , sulfurous ion |
- ___ 12. The nonmetals in Groups 6A and 7A ____.
- lose electrons when they form ions
 - have a numerical charge that is found by subtracting 8 from the group number
 - all have ions with a -1 charge
 - end in *-ite*
- ___ 13. Which of the following is NOT a cation?
- | | |
|------------------|------------------|
| a. iron(III) ion | c. Ca^{2+} |
| b. sulfate | d. mercurous ion |

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N Noddings



Chapter 9 Practice Test Naming And Writing Chemical Formulas:

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web jun 29 2021 environment related performance can also be found in the report in 2018 samsung committed to source renewable energy for 100 of all worksites in the u s china and europe by 2020 with implemented action plans tailored to each region the company achieved its renewable energy target for these worksites in 2020

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web as a multinational company samsung trades globally therefore various factors contribute to the performance of the brand samsung pestle analysis of samsung the pestle analysis is a powerful commercial tool that highlights the influence of various factors on the brand or business

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web when formulating an international business strategy samsung should take external environmental factors into consideration there are various theoretical tools that help analyzing the external environment such as pest analysis five forces analysis and external factor evaluation matrix

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web may 18 2020 pestle analysis of samsung provides visibility into the impact performance and control of businesses also through the macro environmental evaluation political factors samsung is a technology corporation that has to deal with the current that pre existing regulatory legislation

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web growing environmental and socioeconomic risks coupled with geopolitical uncertainties have reinforced our belief that sustainability needs to be a key force for driving our competitiveness and technological innovation to accelerate this drive samsung electronics announced the new environmental strategy on september 15th 2022 and cemented our

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