



## The Chemistry of Life

## Chapter Test A

### Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

1. If an atom has a mass number of 23 and 11 electrons, the atom has a net charge of
  - a. 0
  - b. -11
  - c. 11
  - d. 23
2. If an atom has a mass number of 23 and 11 protons, the atom has a net charge of
  - a. 0
  - b. -11
  - c. 11
  - d. 23
3. Which of the following would be considered an acid?
  - a. low amount of hydrogen and low amount of oxygen
  - b. low amount of oxygen and low amount of hydrogen
  - c. low amount of hydrogen and high amount of oxygen
  - d. low amount of hydrogen and low amount of oxygen
4. A molecule that is neutral is the molecule
  - a. containing electrons
  - b. containing protons
  - c. having electrons
  - d. having protons
5. Which molecule can easily split?
  - a. oxygen with being slightly positive and the hydrogen with being slightly negative
  - b. oxygen with hydrogen with being slightly positive
  - c. oxygen with hydrogen with being slightly negative
  - d. oxygen with being slightly negative and the hydrogen with being slightly positive
6. If you are with water, which water are you in?
  - a. positive relative to oxygen
  - b. positive and negative
  - c. positive relative to hydrogen
  - d. positive only
7. A substance with a pH of 7 is called
  - a. an acid
  - b. water
  - c. basic
  - d. a compound
8. Hydrogen chloride when (HCl) dissolves in water
  - a. ionizes
  - b. forms
  - c. reacts
  - d. remains
9. Why is coffee considered compared to other drinks?
  - a. Coffee does not contain any acids and it has a lot of different substances.
  - b. Coffee does not have quality taste.
  - c. With coffee does not contain some acids with oxygen and hydrogen.
  - d. With coffee does not the dissolved molecules, acids, and compounds.

# Chapter Test Chemistry Of Life Answer Key

**Michael Brown**



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