

# Chapter - 1

## Chemical Reactions and Equations

Chemical Reaction: • Chemical reaction is a process in which new substances with different properties are formed.

- Atoms of one element don't change into another element.
- Only rearrangement of atoms takes place in a chemical reaction.

Examples 1. Hydrogen + Oxygen  $\rightarrow$  water

2. Iron + Oxygen  $\rightarrow$  rust

3. Potassium and chlorine gas  $\rightarrow$  chloride

Physical Change: • No new substance is formed.

- It is a temporary change.
- No change in the composition of the substance takes place.
- It is easily reversible.
- Heat or light may or may not be given out or absorbed.

Chemical Change: • A new substance is formed.

- It is a permanent change.
- The composition of the new substance change greatly.
- The change may be reversible or irreversible.
- Heat or light or both are generally given out or absorbed.

## Law of Conservation of mass

Matter can neither be created nor destroyed in a chemical reaction. This means that total mass of all reactants must

# Chemistry Chapter 1 Notes

**Rosina Ehmman**



## Chemistry Chapter 1 Notes:

Forensic Chemistry Michael Grossman, 2021-12-20 FORENSIC CHEMISTRY FUNDAMENTALS strives to help scientists workplace health safety and other areas are included And two issues of Canadian legal process are argued as essays in the final two chapters

**Notes on books** Longmans, Green and co, 1903 **Notes on Joyce's Ulysses, Part 1 (Chapter 1-3).** Harry Vreeswijk, 1971 The First Miracle Drugs John E. Lesch, 2007 In the decade from 1935-1945 while the Second World War raged in Europe a new class of medicines capable of controlling bacterial infections launched a therapeutic revolution that continues today The new medicines were not penicillin and antibiotics but sulfonamides or sulfa drugs The sulfa drugs preceded penicillin by almost a decade and during World War II they carried the main therapeutic burden in both military and civilian medicine Their success stimulated a rapid expansion of research and production in the international pharmaceutical industry raised expectations of medicine and accelerated the appearance of new and powerful medicines based on research The latter development created new regulatory dilemmas and unanticipated therapeutic problems The sulfa drugs also proved extraordinarily fruitful as starting points for new drugs or classes of drugs both for bacterial infections and for a number of important non-infectious diseases This book examines this breakthrough in medicine pharmacy and science in three parts Part I shows that an industrial research setting was crucial to the success of the revolution in therapeutics that emerged from medicinal chemistry Part II shows how national differences shaped the reception of the sulfa drugs in Germany France Britain and the United States The author uses press coverage of the day to explore popular perceptions of the dramatic changes taking place in medicine Part III documents the impact of the sulfa drugs on the American effort in World War II It also shows how researchers came to an understanding of how the sulfa drugs worked adding a new theoretical dimension to the science of pharmacology and at the same time providing a basis for the discovery of new medicinal drugs in the 1940s 1950s and 1960s A concluding chapter summarizes the transforming impact of the sulfa drugs on twentieth century medicine tracing the therapeutic revolution from the initial breakthrough in the 1930s to the current search for effective treatments for AIDS and the new horizons opened up by the human genome project and stem cell research

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