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# HEAT TRANSFER AND THERMODYNAMICS EXAM QUESTIONS WITH COMPLETE SOLUTIONS

**Radiation** - ANSWER-transfer of energy where there is no conductive medium (for example in space) for heat to transfer through. Heat takes place instead in the form of electromagnetic waves. When these waves fall on an object, some of the energy is absorbed, increasing the objects internal energy.

**Conduction** - ANSWER-heat source and heat sink are connected by matter. Transfer of heat occurs through collisions. Heat energy is transferred in solids, liquids, and gases (but in a lesser extent). Energy is transported by conduction as molecules vibrate, rotate, and/or collide into each other.

**Transfer of heat energy to a mass** - ANSWER-changes its temperature and its dimensions.

**Specific heat is the amount of heat** - ANSWER-needed to raise one gram of material one degree Celsius.

**1 Calorie of heat will raise** - ANSWER-1 gram of water 1 degree C.

**Land areas warm up more rapidly than** - ANSWER-water areas for the same heat input.

**Latent heat is** - ANSWER-the heat required to change a state of matter (from solid to liquid, or liquid to gas)

The water latent heat of fusion is 80 cal/g. The water latent heat of evaporation is 600 cal/g. When water evaporates, it takes heat from the environment. When it condenses, - ANSWER-it releases heat to the environment. Latent heat of evaporation/condensation is an important sink/source of atmospheric energy. This latent heat drives hurricanes and thunderstorms.

**Differences in ground temperatures causes hot and cool spots.** Warm air is forced upwards by the cooler air (convection). Rising air parcel goes to lower pressure. Air parcel expands and cools (gas law). If the air parcel is still warmer (buoyant) than the environment, - ANSWER-then it will continue to rise. On the other hand, if the air parcel is the same or cooler temperature than the environment, then it will stop rising.

**What are methods of heat transfer** - ANSWER-conduction, convection, and radiation

**Conduction** - ANSWER-transfer of energy through matter from one atom/molecules to the next within a substance. Conduction is most effective in solids, but can also take

# Heat Transfer Exam Solutions

**Mike Jess**



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the end of each chapter to test the understanding gained within each subject A solutions manual and PowerPoint slides accompany the text upon qualification

**Heat Transfer Applications for the Practicing Engineer** Louis Theodore, 2011-11-01 This book serves as a training tool for individuals in industry and academia involved with heat transfer applications Although the literature is inundated with texts emphasizing theory and theoretical derivations the goal of this book is to present the subject of heat transfer from a strictly pragmatic point of view The book is divided into four Parts Introduction Principles Equipment Design Procedures and Applications and ABET related Topics The first Part provides a series of chapters concerned with introductory topics that are required when solving most engineering problems including those in heat transfer The second Part of the book is concerned with heat transfer principles Topics that receive treatment include Steady state Heat Conduction Unsteady state Heat Conduction Forced Convection Free Convection Radiation Boiling and Condensation and Cryogenics Part three considered the heart of the book addresses heat transfer equipment design procedures and applications In addition to providing a detailed treatment of the various types of heat exchangers this part also examines the impact of entropy calculations on exchanger design and operation maintenance and inspection OM I plus refractory and insulation effects The concluding Part of the text examines ABET Accreditation Board for Engineering and Technology related topics of concern including economics and finance numerical methods open ended problems ethics environmental management and safety and accident management

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