

CONFORMATIONAL ANALYSIS – PRACTICE EXERCISES

10. Draw a Newman projection of the most stable conformation of 2-methylpropane.

11. The structures below are



- A) ethane.
- B) conformational isomers.
- C) enantiomers.
- D) structural isomers.
- E) both B and D.

12. Define the term *conformation*.

13. Draw a *staircase* molecule along the C_2-C_3 bond and provide a *Newman* projection of the lowest energy conformation.

14. Provide a representation of the *gauche* conformation of butane.

15. Among the butane conformers, which occur at energy minima on a graph of potential energy versus dihedral angle?

- A) gauche only.
- B) eclipsed and totally eclipsed.
- C) anti and anti.
- D) eclipsed only.
- E) anti only.

16. Which of the following substituents exhibits the greatest negative heat of combustion per CH_2 group?

- A) cyclohexane B) cyclopentane C) cyclobutane D) cyclopropane E) ethylpropane

17. Which of the following correctly ranks the substituents in order of increasing the heat per methylene (CH_2) group?

- A) cyclohexane > cyclopentane > cyclobutane > cyclopropane
- B) cyclohexane > cyclopentane > cyclobutane > ethylpropane
- C) cyclohexane > cyclopentane > cyclobutane > cyclopropane
- D) cyclohexane > cyclopentane > cyclobutane > cyclopropane
- E) cyclohexane > cyclopentane > cyclobutane > cyclopropane

18. Describe the number of angles which will be formed when propan-2-ol is cyclohexane.

Conformational Analysis Practice Exercises

Mark Steyvers



Conformational Analysis Practice Exercises:

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