

STOICHIOMETRY LIMITING REAGENT

- ① Consider the following reaction:



Balance the reaction.

- a 200 grams of each reactant react. How many grams of silver phosphate are formed?

- b How much excess reagent is left?

- ② Consider the following reaction:



Balance the reaction.

How many grams of NH_3 are produced when 28 g of N_2 reacts with 25 g of H_2 ? What is the limiting reagent? What is the excess reagent? How many grams of excess reagent remain?

Stoichiometry Limiting Reagent Problems Answers

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