

## VOLUMETRIC ANALYSIS -CLASS XI

### EXPERIMENT-1

#### Aim:

To prepare 100ml  $\frac{M}{20}$  sodium carbonate solution and determine the Molarity and the strength of the given solution of hydrochloric acid using  $\frac{M}{20}$  sodium carbonate solution.

#### Apparatus required:

Burette, Pipette, Burette stand, conical flask.

#### Chemicals Required:

Sodium carbonate, hydrochloric acid, Indicator (methyl orange), distilled water.

#### Theory:

- To prepare 100ml  $\frac{M}{20}$  sodium carbonate solution

Weight of  $\text{Na}_2\text{CO}_3$  required for preparing 100 ml  $M/20$   $\text{Na}_2\text{CO}_3$  solution.

$$\frac{M}{20} = \frac{\text{weight of sodium carbonate required for 100 ml} \times 1000}{\text{molar mass of sodium carbonate} \times 100}$$

$$\therefore \frac{1}{20} = \frac{x \times 1000}{106 \times 100} \Rightarrow \frac{x \times 10}{106}$$

$$\frac{1}{20} = \frac{10x}{106}$$

$$= \frac{106}{200} = 0.53 \text{ gm.}$$

# Xi Chemistry Practical Procedure Volumetric Analysis

**SA Dillow**



## **Xi Chemistry Practical Procedure Volumetric Analysis:**

## **Xi Chemistry Practical Procedure Volumetric Analysis** Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Xi Chemistry Practical Procedure Volumetric Analysis**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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