

8E Combustion- Revision Worksheet

Define the following keywords:

Combustion	
Fuel	
Fossil Fuel	
Hydrocarbon	
Reactant	
Product	

Describe how hydrogen can be used to power vehicles:

.....

.....

Write out a word equation for the reaction in a fuel cell:

Describe the energy transfers that occur during combustion.

.....

.....

State the products of hydrocarbon combustion and explain how they are formed.

.....

.....

.....

.....

Explain how the equipment here allows you to test for the products of combustion.

.....

.....

Complete the following word equations:

Magnesium + Oxygen →

Iron + Oxygen →

Copper + Oxygen →

Zinc + Oxygen →

Explain why the reactions above are examples of oxidation.

.....

.....

Describe what is meant by the 'law of conservation of mass'

.....

.....

5g of magnesium is burnt in air, producing 10g of magnesium oxide. Calculate how much oxygen was reacted with.

Describe the reaction of magnesium with oxygen.

Explain why the mass of zinc oxide is greater than the mass of the zinc reacted.

.....

Describe why combustion is an exothermic reaction.

.....

.....

Label the fire triangle:

State what each of the hazard symbols means:

Complete the following table about the different fire extinguishers:

Type	Colour	Explanation of what it puts out
Water		
CO2		
Foam		
Powder		

Year 8 Exploring Science Test Answers

JE Gale



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Table of Contents Year 8 Exploring Science Test Answers

1. Understanding the eBook Year 8 Exploring Science Test Answers
 - The Rise of Digital Reading Year 8 Exploring Science Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Year 8 Exploring Science Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Year 8 Exploring Science Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Year 8 Exploring Science Test Answers
 - Personalized Recommendations
 - Year 8 Exploring Science Test Answers User Reviews and Ratings
 - Year 8 Exploring Science Test Answers and Bestseller Lists
5. Accessing Year 8 Exploring Science Test Answers Free and Paid eBooks
 - Year 8 Exploring Science Test Answers Public Domain eBooks
 - Year 8 Exploring Science Test Answers eBook Subscription Services
 - Year 8 Exploring Science Test Answers Budget-Friendly Options

6. Navigating Year 8 Exploring Science Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Year 8 Exploring Science Test Answers Compatibility with Devices
 - Year 8 Exploring Science Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Year 8 Exploring Science Test Answers
 - Highlighting and Note-Taking Year 8 Exploring Science Test Answers
 - Interactive Elements Year 8 Exploring Science Test Answers
8. Staying Engaged with Year 8 Exploring Science Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Year 8 Exploring Science Test Answers
9. Balancing eBooks and Physical Books Year 8 Exploring Science Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Year 8 Exploring Science Test Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Year 8 Exploring Science Test Answers
 - Setting Reading Goals Year 8 Exploring Science Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Year 8 Exploring Science Test Answers
 - Fact-Checking eBook Content of Year 8 Exploring Science Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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