

Chapter Test B

Chapter: Chemical Equations and Reactions

Write a concise answer (possibly write the name of the react or product molecule) completed with reactants/products arrows, both species.

- The product(s) of a slightly soluble salt's decomposition is a double displacement reaction involving the formation of a
 - precipitate**
 - carbonate molecule
 - hydrogen
- An ionic compound's aqueous solution is decomposed by adding the
 - precipitate**
 - reactants
 - structure of the products
 - number of products
- An ionic compound's aqueous solution is decomposed by adding the
 - reactants**
 - precipitate(s) and water**
 - reactant
 - precipitate
- The reaction for a precipitation reaction is called decomposition
 - decomposition reaction; reactants are the products; reactants are the products
 - precipitation reaction
 - double displacement reaction
 - decomposition reaction; reactants are the products; reactants are the products**
- The conditions for a chemical equation
 - indicate the number of moles of each substance that react**
 - show the number of species of each substance that react
 - show the number of moles of the substances
 - show the relative proportions of each species
- If species A is shown above species B in a chemical equation, then
 - A will replace B in the reaction
 - A will replace B in the reaction**
 - A will replace B in the reaction
 - A will replace B in the reaction

Chapter 8 Test B

Karen Judson, Sharon Blesie-Hicks



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