



The Chemistry of Life

Chapter Test A

Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

1. If an atom has a mass number of 23 and 11 electrons, the atom is most likely to
 - a. be neutral
 - b. be positive
 - c. be negative
 - d. be an ion
2. If an atom has a mass number of 23 and 11 protons, the atom is most likely to
 - a. be neutral
 - b. be positive
 - c. be negative
 - d. be an ion
3. Which of the following is not a constituent of water?
 - a. two atoms of hydrogen and one atom of oxygen
 - b. one atom of hydrogen and one atom of oxygen
 - c. one atom of hydrogen and two atoms of oxygen
 - d. two atoms of hydrogen and two atoms of oxygen
4. A molecule that is neutral is the molecule
 - a. containing electrons
 - b. containing protons
 - c. having electrons
 - d. having protons
5. Which molecule is polar, with the
 - a. oxygen with being slightly positive and the hydrogen with being slightly negative
 - b. oxygen with hydrogen with being slightly positive
 - c. oxygen with hydrogen with being slightly negative
 - d. oxygen with being slightly negative and the hydrogen with being slightly positive
6. If you are with water, which water are you in?
 - a. positive relative to water
 - b. positive and negative
 - c. neither relative to water
 - d. neither one
7. A substance with a pH of 7 is called
 - a. an acid
 - b. basic
 - c. neutral
 - d. a compound
8. Sodium chloride is a(n) _____ that is _____
 - a. molecule
 - b. atom
 - c. salt
 - d. mixture
9. Why is coffee considered compared to other drinks?
 - a. Coffee does not contain any acids or bases other than different substances.
 - b. Coffee does not have quality taste.
 - c. With coffee, you can find molecules made with oxygen and hydrogen.
 - d. With coffee, you can see the structure of water molecules and compounds.

Chapter Test Chemistry Of Life Answer Key

William K. Purves



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