

CHEMISTRY MULTIPLE CHOICE QUESTIONS

A. Atoms, molecules and stoichiometry

2002 -2019



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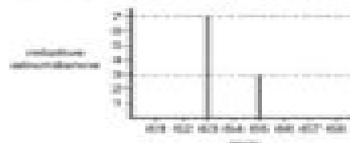
As a combustion, an alkane burns can be considered to have a fully set of 1.00 kg of methane (empirical formula CH_4).

Which mass of carbon dioxide does a person produce each day if all the carbon in their is digested and oxidised?

- A 0.007 kg B 0.006 kg C 1.30 kg D 0.04 kg

(2002 NA/12)

The diagram shows the mass spectrum of a sample of naturally-occurring copper.



What is the relative atomic mass of this copper?

- A 63.2 B 63.5 C 63.8 D 64.0

(2002 NA/12)

Which pairs of compounds have the same empirical formula?

- ethane and ethene
- ethane and cyclohexane
- cyclohexane and oct-1-ene

(2002 NA/12)

A stream of 10cm^3 of methane and 10cm^3 of ethane was sparked with an excess of oxygen. After cooling to room temperature, the residual gas was passed through aqueous potassium hydroxide.

Which compounds have the empirical formula CH_2O ?

- methanol
- ethanoic acid
- methyl methanoate

(2002 NA/12)

What is the number of molecules in 500 cm^3 of oxygen under room conditions?

- A 1.25×10^{23}
- B 1.25×10^{24}
- C 3.0×10^{23}
- D 3.0×10^{24}

(2004 NA/12)

In the preparation of soft margarine, glycerol trioleate



is fully hydrogenated so that, on average, one of its side-chains is converted into the $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$ moiety and two side-chains are converted into the $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$ moiety.

How many moles of hydrogen are required to convert one mole of glycerol trioleate into the soft margarine?

- A 4 B 5 C 6 D 8

(2008 NA/12)

Use of the Data Booklet is relevant to this question.

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of electrolysis reactions conductors and non conductors dry cells electrical devices circuit symbols electrolytes non electrolytes organic solvents polarization and valence electrons The chapter Electricity and Chemicals MCQs covers topics of chemical to electrical energy dry cells electrolyte non electrolyte and polarization The chapter Elements Compounds and Mixtures MCQs covers topics of elements compounds mixtures molecules atoms and symbols for elements *Oswaal NTA CUET (UG) | Question Bank Chapterwise & Topicwise Chemistry For 2024 Exam* Oswaal Editorial Board,2023-11-29 Description of the product Strictly as per the Latest Exam Pattern issued by NTA 100% Updated with 2023 Exam Paper Previous Years Questions 2021 2023 for better Exam insights Revision Notes for Crisp Revision with Smart Mind Maps Concept Videos for complex concepts clarity 800 Questions for Extensive Practice **Oswaal NTA CUET (UG) Question Banks | Chapterwise & Topicwise | English, Physics, Chemistry, Math & General Test | Set of 5 Books | Entrance Exam Preparation Books 2024** Oswaal Editorial Board,2024-03-08 Description of the Product 100% Exam Ready With 2023 CUET UG Exam Papers Fully Solved with Explanations Concept Clarity With Revision Notes Chapter Analysis with updated pattern Extensive Practice With 800 Practice Questions of Previous Years 2021 2023 Fill Learning Gaps with Smart Mind Maps Concept Videos Valuable Exam Insights With Tips Tricks to ace CUET UG in 1st Attempt *Excel HSC & Preliminary Senior Science* Jennifer Hill,2011 This comprehensive study guide covers the complete HSC Preliminary Senior Science course and has been specifically created to maximise exam success This guide has been designed to meet all study needs providing up to date information in an easy to use format The sample HSC Exam has been updated for the new format Excel HSC Preliminary Senior Science contains an introductory section including how to use the book and an explanation of the new course helpful study and exam techniques comprehensive coverage of the entire Preliminary and HSC courses hundreds of diagrams to aid understanding icons and boxes to highlight key concepts and assessment skills including laboratory and field work checklists of key terms end of chapter revision questions with fully explained answers a trial HSC style exam with answers and explanations a glossary of key terms useful websites highlighted throughout Grade 10 Chemistry Multiple Choice Questions and Answers (MCQs) Arshad Iqbal,2020-03-13 Grade 10 Chemistry Multiple Choice Questions and Answers MCQs Grade 10 chemistry quizzes practice tests with answer key provides mock tests for competitive exams to solve 842 MCQs Grade 10 Chemistry MCQs helps with theoretical conceptual and analytical study for self assessment career tests This book can help to learn and practice Grade 10 Chemistry quizzes as a quick study guide for placement test preparation Grade 10 Chemistry Multiple Choice Questions and Answers MCQs is a revision guide with a collection of trivia quiz questions and answers on topics Acids bases and salts biochemistry characteristics of acids bases and salts chemical equilibrium chemical industries environmental chemistry atmosphere water hydrocarbons and organic chemistry to enhance teaching and learning Grade 10 Chemistry Quiz Questions and Answers pdf also covers the syllabus of many competitive papers for admission exams of different schools from chemistry textbooks on chapters Acids Bases and

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